

ELECTROPHOTOGRAPHIC PHOTSENSITIVE MEMBER,
PROCESS CARTRIDGE AND ELECTROPHOTOGRAPHIC APPARATUS

BACKGROUND OF THE INVENTION

5 Field of the Invention

This invention relates to an electrophotographic photosensitive member, a process cartridge and an electrophotographic apparatus. More particularly, it relates to an electrophotographic photosensitive member
10 having in its photosensitive layer a specific charge-transporting material, and a process cartridge and an electrophotographic apparatus which have the electrophotographic photosensitive member.

Related Background Art

15 Image-bearing members used in electrophotographic photosensitive member include electrophotographic photosensitive members. Form the viewpoints of high productivity and future developments and easiness of material designing, organic electrophotographic
20 photosensitive members making use of organic photoconductive materials are energetically put forward. In regard to the functionality as electrophotographic photosensitive members, electrophotographic photosensitive members superior to inorganic
25 electrophotographic photosensitive member have come to be manufactured. It, however, is desired to achieve higher sensitivity and to improve image stability in repeated

use and durability in organic electrophotographic photosensitive members.

To settle these subjects, approaches to the improvement of charge-transporting materials have been
5 proposed (see, e.g., Japanese Patent Application Laid-open Nos. H9-292724 and 2001-133994). Japanese Patent Application Laid-open Nos. H9-292724 and 2001-133994 disclose that a charge-transporting material having three or four structures having in its molecule a
10 tertiary amine in which phenyl groups are bonded to the nitrogen atom (hereinafter "triphenylamine structures") is used in an electrophotographic photosensitive member to achieve a higher sensitivity, and superior charge transport performance is brought out by the three or four
15 triphenylamine structures possessed in the charge-transporting material. There, however, is no disclosure as to a charge-transporting material having five or more structures having a tertiary amine in which either of substituted or unsubstituted aromatic
20 carbocyclic groups and substituted or unsubstituted aromatic heterocyclic groups are bonded to the nitrogen atom (hereinafter "triarylamine structures") charge-transporting material. There is also no disclosure as to the durability in mechanical strength of the
25 electrophotographic photosensitive member.

Similarly, Japanese Patent Application Laid-open No. 2000-206721 gives an example of a charge-transporting

material having therein two to four triphenylamine structures. This Japanese Patent Application Laid-open No. 2000-206721 discloses that the charge-transporting material having two to four triphenylamine structures
5 makes the charge transport layer have higher glass transition temperature, but admits in Comparative Examples that only the incorporation of this charge-transporting material can not achieve the improvement in durability.

10 Thus, the improvement of charge-transporting materials have achieved a superior charge transport performance. Since, however, a low-molecular-weight charge-transporting material is mixed in a binder resin, it would be said that the inherent mechanical strength of
15 the binder resin is not necessarily fully exhibited.

For the purpose of more preferably preventing the layer from having a low mechanical strength because of the addition of a low-molecular-weight charge-transporting material, Japanese Patent Application
20 Laid-open No. S61-151545 and Japanese Patent Publication No. H5-49106 disclose a high-molecular-weight charge-transporting material of a polymer type which contains triarylamine structures in a large number. In these patent publications, it is disclosed that the
25 high-molecular-weight charge-transporting material of a polymer type which contains triarylamine structures in a large number is used in an electrophotographic

photosensitive member, thereby improving its durability in repeated use. According to the disclosure in these patent publications, the high-molecular-weight charge-transporting material is synthesized by
5 polymerization reaction, and hence it is produced in the form of a mixture containing charge-transporting materials with various molecular weights. However, in Japanese Patent Publication No. H5-49106, it is also disclosed that there is little difference from a
10 high-molecular-weight charge-transporting material having no molecular weight distribution as long as the number of times of the repetition of repeating structural units is 10 times or less.

As examples in which a high-molecular-weight
15 charge-transporting material of a polymer type is similarly used in an electrophotographic photosensitive member, those disclosed in International Publications No. WO00/078843 and No. WO99/32537 may be given. A high-molecular-weight charge-transporting material
20 disclosed in International Publication No. WO00/078843 is a high-molecular-weight charge-transporting material of a polymer type having a molecular weight distribution, which is produced by polymerization reaction of a low-molecular-weight monomer, showing that the use of
25 this charge-transporting material in the electrophotographic photosensitive member enables higher sensitivity to be achieved in virtue of an improvement in

durability and an improvement in charge transport performance.

Publication No. WO00/078843 further discloses a method in which a high-molecular-weight
5 charge-transporting material of a polymer type which contains triphenylamine structures in a large number is separated into molecular-weight fractions, and also discloses that such separation into molecular-weight
10 fractions brings an improvement in charge transport performance. International Publication No. WO99/32537 further discloses that the use of a high-molecular-weight charge-transporting material of a polymer type brings an improvement in durability of electrophotographic
15 photosensitive members, proposing an electrophotographic photosensitive member having a high charge transport performance and a high durability.

However, even if the polymer-type charge-transporting material has a small number of times of the repetition of repeating structural units and is
20 separated into molecular-weight fractions, it has a molecular weight distribution and contains charge-transporting materials with various molecular weights. Hence, all the polymer-type high-molecular-weight charge-transporting materials do
25 not necessarily have sufficient mechanical strength and electrophotographic performance. Also, even when they have a certain mechanical strength, they have such a

disadvantage that their manufacturing cost is so high as to be not suited for practical use.

In addition, since an improvement in wear resistance makes the depth of wear smaller, the
5 photosensitive layer have a longer lifetime and the influence of electrical external force coming from the steps of charging, imagewise exposure, development with toner and transfer the photosensitive layer undergoes, becomes relatively large. Hence, when the photosensitive
10 layer is repeatedly used, such influence is liable to appear. For example, smeared images may appear in a high-humidity environment due to deterioration in the electrophotographic photosensitive member surface. Thus, problems have come to arise which should be solved at the
15 same time the durability of electrophotographic photosensitive members is improved.

SUMMARY OF THE INVENTION

An object of the present invention is to provide an
20 electrophotographic photosensitive member which has solved the problems discussed above, and has high mechanical strength such as wear resistance and scratch resistance and superior stability in repeated use in virtue of the incorporation of a binder resin and a
25 specific charge-transporting material in a photosensitive layer.

Another object of the present invention is to

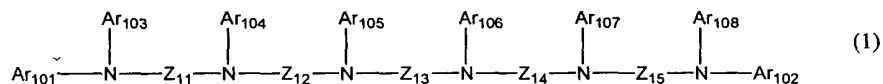
provide a process cartridge and an electrophotographic apparatus which have such an electrophotographic photosensitive member.

That is, the present invention provides an
5 electrophotographic photosensitive member comprising a support and a photosensitive layer provided on the support, wherein;

the photosensitive layer contains one or two or more kind(s) of charge-transporting material(s);

10 at least one kind of charge-transporting material contained in the photosensitive layer is a charge-transporting material having a structure represented by the following Formula (1) and having a molecular weight of from 1,500 to 4,000; and

15 the charge-transporting material having a structure represented by the following Formula (1) and having a molecular weight of from 1,500 to 4,000 is held in a proportion of from 90% by weight to 100% by weight based on the total weight of the charge-transporting
20 material(s) contained in the photosensitive layer:



wherein Ar₁₀₁ to Ar₁₀₈ each independently represent a substituted or unsubstituted monovalent aromatic carbocyclic group or a substituted or unsubstituted
25 monovalent aromatic heterocyclic group, and Z₁₁ to Z₁₅ each independently represent a substituted or unsubstituted

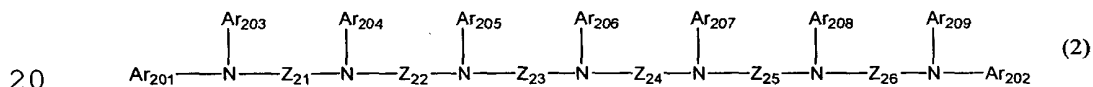
divalent aromatic carbocyclic group or a substituted or unsubstituted divalent aromatic heterocyclic group.

The present invention also provides an electrophotographic photosensitive member comprising a support and a photosensitive layer provided on the support, wherein;

the photosensitive layer contains one or two or more kind(s) of charge-transporting material(s);

at least one kind of charge-transporting material contained in the photosensitive layer is a charge-transporting material having a structure represented by the following Formula (2) and having a molecular weight of from 1,500 to 4,000; and

the charge-transporting material having a structure represented by the following Formula (2) and having a molecular weight of from 1,500 to 4,000 is held in a proportion of from 90% by weight to 100% by weight based on the total weight of the charge-transporting material(s) contained in the photosensitive layer:



wherein Ar₂₀₁ to Ar₂₀₉ each independently represent a substituted or unsubstituted monovalent aromatic carbocyclic group or a substituted or unsubstituted monovalent aromatic heterocyclic group, and Z₂₁ to Z₂₆ each independently represent a substituted or unsubstituted divalent aromatic carbocyclic group or a substituted or

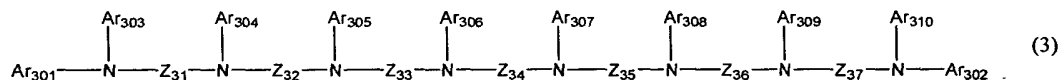
unsubstituted divalent aromatic heterocyclic group.

The present invention still also provides an electrophotographic photosensitive member comprising a support and a photosensitive layer provided on the support, wherein;

the photosensitive layer contains one or two or more kind(s) of charge-transporting material(s);

at least one kind of charge-transporting material contained in the photosensitive layer is a charge-transporting material having a structure represented by the following Formula (3) and having a molecular weight of from 1,500 to 4,000; and

the charge-transporting material having a structure represented by the following Formula (3) and having a molecular weight of from 1,500 to 4,000 is held in a proportion of from 90% by weight to 100% by weight based on the total weight of the charge-transporting material(s) contained in the photosensitive layer:



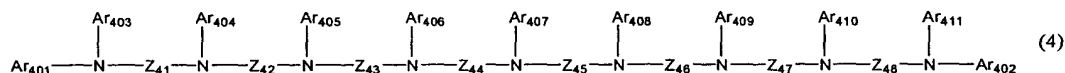
wherein Ar₃₀₁ to Ar₃₁₀ each independently represent a substituted or unsubstituted monovalent aromatic carbocyclic group or a substituted or unsubstituted monovalent aromatic heterocyclic group, and Z₃₁ to Z₃₇ each independently represent a substituted or unsubstituted divalent aromatic carbocyclic group or a substituted or unsubstituted divalent aromatic heterocyclic group.

The present invention further provides an electrophotographic photosensitive member comprising a support and a photosensitive layer provided on the support, wherein;

5 the photosensitive layer contains one or two or more kind(s) of charge-transporting material(s);

at least one kind of charge-transporting material contained in the photosensitive layer is a charge-transporting material having a structure
10 represented by the following Formula (4) and having a molecular weight of from 1,500 to 4,000; and

the charge-transporting material having a structure represented by the following Formula (4) and having a molecular weight of from 1,500 to 4,000 is held in a
15 proportion of from 90% by weight to 100% by weight based on the total weight of the charge-transporting material(s) contained in the photosensitive layer:



wherein Ar₄₀₁ to Ar₄₁₁ each independently represent a
20 substituted or unsubstituted monovalent aromatic carbocyclic group or a substituted or unsubstituted monovalent aromatic heterocyclic group, and Z₄₁ to Z₄₈ each independently represent a substituted or unsubstituted divalent aromatic carbocyclic group or a substituted or
25 unsubstituted divalent aromatic heterocyclic group.

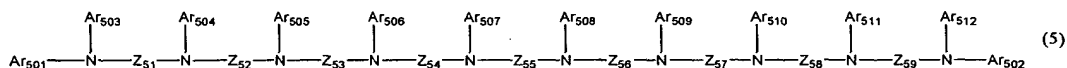
The present invention still further provides an

electrophotographic photosensitive member comprising a support and a photosensitive layer provided on the support, wherein;

the photosensitive layer contains one or two or
5 more kind(s) of charge-transporting material(s);

at least one kind of charge-transporting material contained in the photosensitive layer is a charge-transporting material having a structure represented by the following Formula (5) and having a
10 molecular weight of from 1,500 to 4,000; and

the charge-transporting material having a structure represented by the following Formula (5) and having a molecular weight of from 1,500 to 4,000 is held in a proportion of from 90% by weight to 100% by weight based
15 on the total weight of the charge-transporting material(s) contained in the photosensitive layer:



wherein Ar₅₀₁ to Ar₅₁₂ each independently represent a substituted or unsubstituted monovalent aromatic
20 carbocyclic group or a substituted or unsubstituted monovalent aromatic heterocyclic group, and Z₅₁ to Z₅₉ each independently represent a substituted or unsubstituted divalent aromatic carbocyclic group or a substituted or unsubstituted divalent aromatic heterocyclic group.

25 The present invention still further provides a process cartridge which comprises any one of the

electrophotographic photosensitive members described
above and at least one means selected from the group
consisting of a charging means, a developing means and a
cleaning means, which are integrally supported; and is
5 detachably mountable on the main body of an
electrophotographic apparatus.

The present invention still further provides an
electrophotographic apparatus comprising any one of the
electrophotographic photosensitive members described
10 above, a charging means, an exposure means, a developing
means and a transfer means.

BRIEF DESCRIPTION OF THE DRAWINGS

Fig. 1 shows an MALDI-TOF-MASS spectrum of an
15 exemplary compound (CT-10) of the present invention.

Fig. 2 shows an MALDI-TOF-MASS spectrum of an
exemplary compound (CT-17) of the present invention.

Fig. 3 shows an MALDI-TOF-MASS spectrum of an
exemplary compound (CT-39) of the present invention.

20 Fig. 4 is a schematic view showing an example of
the construction of an electrophotographic apparatus
having the electrophotographic photosensitive member of
the present invention.

25 DESCRIPTION OF PREFERRED EMBODIMENTS

The present invention is described below in detail.

The electrophotographic photosensitive member of

the present invention has a support and a photosensitive layer provided on the support, and the photosensitive layer contains a charge-transporting material.

5 The charge-transporting material in the present invention refers to any of high-molecular-weight charge-transporting materials which can be represented by specific chemical structural formulas.

The charge-transporting material contained in the photosensitive layer is a charge-transporting material
10 having any of structures represented by the above Formulas (1) to (5) and having a molecular weight of from 1,500 to 4,000, and in addition thereto may simultaneously contain a low-molecular-weight charge-transporting material(s) and/or other
15 high-molecular-weight charge-transporting material(s). From the viewpoint of mechanical strength and electrophotographic performance of the electrophotographic photosensitive member, preferred is a charge-transporting material in which:

20 the charge-transporting material having a structure represented by the above Formula (1) and having a molecular weight of from 1,500 to 4,000 is held in a proportion of from 90% by weight to 100% by weight, more preferably from 95% by weight to 100% by weight, and
25 still more preferably 100% by weight, based on the total weight of the charge-transporting material(s) contained in the photosensitive layer;

the charge-transporting material having a structure represented by the above Formula (2) and having a molecular weight of from 1,500 to 4,000 is held in a proportion of from 90% by weight to 100% by weight, more
5 preferably from 95% by weight to 100% by weight, and still more preferably 100% by weight, based on the total weight of the charge-transporting material(s) contained in the photosensitive layer;

the charge-transporting material having a structure
10 represented by the above Formula (3) and having a molecular weight of from 1,500 to 4,000 is held in a proportion of from 90% by weight to 100% by weight, more preferably from 95% by weight to 100% by weight, and still more preferably 100% by weight, based on the total
15 weight of the charge-transporting material(s) contained in the photosensitive layer;

the charge-transporting material having a structure represented by the above Formula (4) and having a molecular weight of from 1,500 to 4,000 is held in a
20 proportion of from 90% by weight to 100% by weight, more preferably from 95% by weight to 100% by weight, and still more preferably 100% by weight, based on the total weight of the charge-transporting material(s) contained in the photosensitive layer; or

25 the charge-transporting material having a structure represented by the above Formula (5) and having a molecular weight of from 1,500 to 4,000 is held in a

proportion of from 90% by weight to 100% by weight, more preferably from 95% by weight to 100% by weight, and still more preferably 100% by weight, based on the total weight of the charge-transporting material(s) contained
5 in the photosensitive layer.

In the above Formula (1), Formula (2), Formula (3), Formula (4) and Formula (5), Ar₁₀₁ to Ar₁₀₈, Ar₂₀₁ to Ar₂₀₉, Ar₃₀₁ to Ar₃₁₀, Ar₄₀₁ to Ar₄₁₁, and Ar₅₀₁ to Ar₅₁₂ are each independently a substituted or unsubstituted monovalent
10 aromatic carbocyclic (hydrocarbon ring) group or a substituted or unsubstituted monovalent aromatic heterocyclic group. The substituted or unsubstituted monovalent aromatic carbocyclic group may include a phenyl group, a naphthyl group, an anthracenyl group and
15 a pyrenyl group, and the substituted or unsubstituted monovalent aromatic heterocyclic group may include a pyridyl group, an indole group, a quinolinyl group, a benzofuranyl group, a dibenzofuranyl group, a benzothiophenyl group and a dibenzothiophenyl group. In
20 particular, a phenyl group, a naphthyl group, a pyridyl group, a benzofuranyl group and a benzothiophenyl group are preferred. The substituents of these groups may include a hydrogen atom, an alkyl group having 1 to 8 carbon atoms, an aromatic carbocyclic group having 3 to
25 12 carbon atoms, an alkoxy group having 1 to 8 carbon atoms, a halogen atom, a fluoroalkyl group, a cyano group and a nitro group. In particular, a hydrogen atom, a

methyl group, an ethyl group, a methoxyl group, a fluorine atom, a chlorine atom, a bromine atom and a trifluoromethyl group are preferred.

In the above Formula (1), Formula (2), Formula (3),
5 Formula (4) and Formula (5), Z_{11} to Z_{15} , Z_{21} to Z_{26} , Z_{31} to Z_{37} , Z_{41} to Z_{48} , and Z_{51} to Z_{59} are each independently a substituted or unsubstituted divalent aromatic carbocyclic group or a substituted or unsubstituted divalent aromatic heterocyclic group. The substituted or
10 unsubstituted divalent aromatic carbocyclic group may include a phenylene group, a biphenylene group, a terphenylene group, a fluorenylene group, a naphthylene group, an anthracenylenylene group and a pyrenylene group, and the substituted or unsubstituted divalent aromatic
15 heterocyclic group may include a pyridinylenylene group, an indolylenylene group, a quinolinylenylene group, a benzofuranylenylene group, a dibenzofuranylenylene group, a benzothiophenylenylene group and a dibenzothiophenylenylene group. The substituted or unsubstituted divalent aromatic carbocyclic group or
20 substituted or unsubstituted divalent aromatic heterocyclic group also may include any of the above substituted or unsubstituted divalent aromatic carbocyclic groups or substituted or unsubstituted divalent aromatic heterocyclic groups formed by bonding
25 through a single bond, a substituted or unsubstituted alkylene group having 1 to 4 carbon atoms, an alkylidene group, a substituted or unsubstituted silylene group

having 1 to 4 silicon atoms, an oxygen atom or a sulfur atom. Of these substituted or unsubstituted divalent aromatic carbocyclic groups or substituted or unsubstituted divalent aromatic heterocyclic groups, a
5 biphenylene group, a fluorenylene group, a pyridinylene group, a dibenzofuranylene group and a benzothiophenylylene group are preferred.

More preferred is that:

Z_{11} to Z_{15} in Formula (1) are each a substituted or
10 unsubstituted biphenylene group, a substituted or unsubstituted dibenzofuranylene group or a substituted or unsubstituted dibenzothiophenylylene group;

Z_{21} to Z_{26} in Formula (2) are each a substituted or
unsubstituted biphenylene group, a substituted or
15 unsubstituted dibenzofuranylene group or a substituted or unsubstituted dibenzothiophenylylene group;

Z_{31} to Z_{37} in Formula (3) are each a substituted or
unsubstituted biphenylene group, a substituted or
unsubstituted dibenzofuranylene group or a substituted or
20 unsubstituted dibenzothiophenylylene group;

Z_{41} to Z_{48} in Formula (4) are each a substituted or
unsubstituted biphenylene group, a substituted or
unsubstituted dibenzofuranylene group or a substituted or
unsubstituted dibenzothiophenylylene group; or

25 Z_{51} to Z_{59} in Formula (5) are each a substituted or
unsubstituted biphenylene group, a substituted or
unsubstituted dibenzofuranylene group or a substituted or

unsubstituted dibenzothiophenylene group.

Still more preferred is that:

of Z_{11} to Z_{15} in Formula (1), one is a substituted or unsubstituted dibenzofuranylene group or a substituted or unsubstituted dibenzothiophenylene, and the others are
5 each a substituted or unsubstituted biphenylene group;

of Z_{21} to Z_{26} in Formula (2), one is a substituted or unsubstituted dibenzofuranylene group or a substituted or unsubstituted dibenzothiophenylene group, and the others
10 are each a substituted or unsubstituted biphenylene group;

of Z_{31} to Z_{37} in Formula (3), one is a substituted or unsubstituted dibenzofuranylene group or a substituted or unsubstituted dibenzothiophenylene group, and the others
15 are each a substituted or unsubstituted biphenylene group;

of Z_{41} to Z_{48} in Formula (4), one is a substituted or unsubstituted dibenzofuranylene group or a substituted or unsubstituted dibenzothiophenylene group, and the others
20 are each a substituted or unsubstituted biphenylene group; or

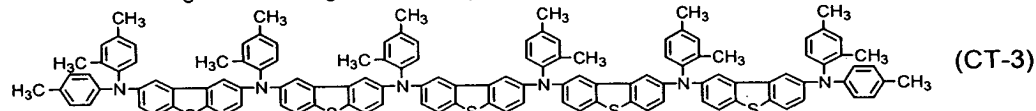
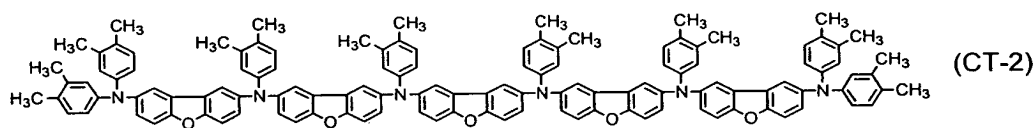
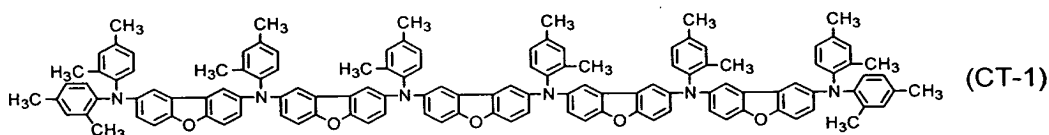
of Z_{51} to Z_{59} in Formula (5), one is a substituted or unsubstituted dibenzofuranylene group or a substituted or unsubstituted dibenzothiophenylene group, and the others
25 are each a substituted or unsubstituted biphenylene group.

The substituents of these substituted or unsubstituted divalent aromatic carbocyclic groups or

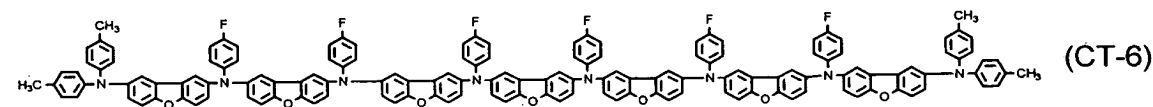
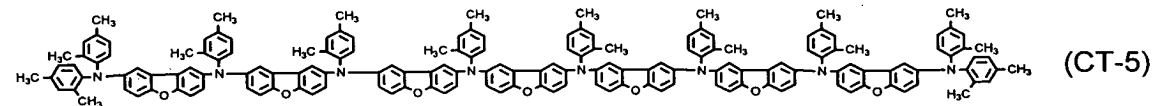
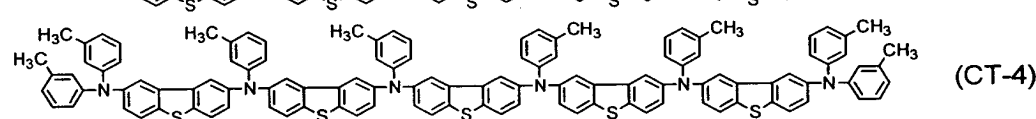
substituted or unsubstituted divalent aromatic heterocyclic groups may include a hydrogen atom, an alkyl group having 1 to 8 carbon atoms, an aromatic carbocyclic group having 3 to 12 carbon atoms, an alkoxyl group

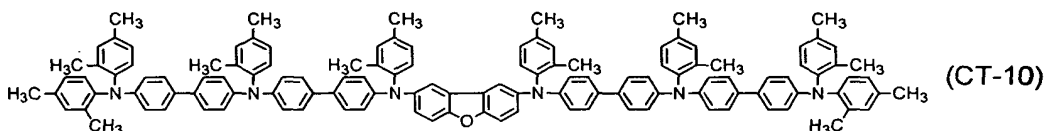
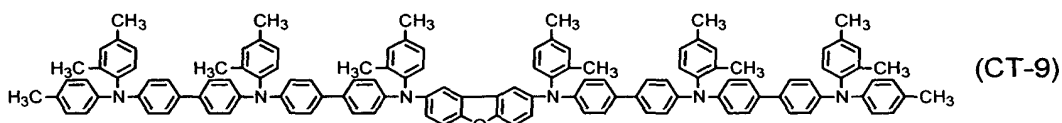
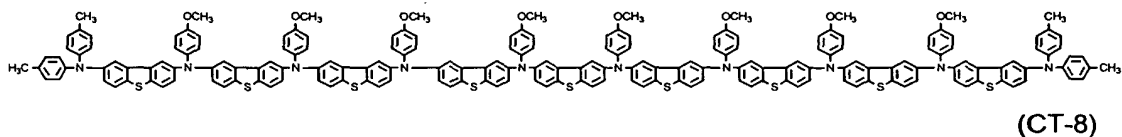
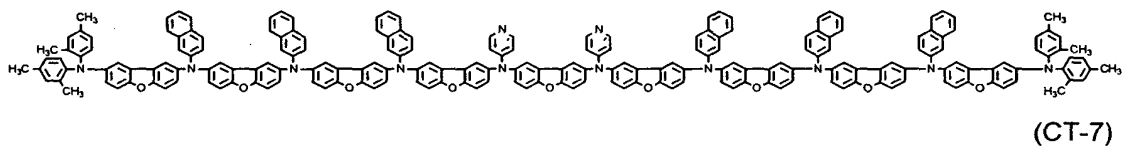
5 having 1 to 8 carbon atoms, a halogen atom, a fluoroalkyl group, a cyano group and a nitro group. In particular, a hydrogen atom, a methyl group, an ethyl group, a methoxyl group, an ethoxyl group, a fluorine atom, a chlorine atom, a bromine atom and a trifluoromethyl group are preferred.

10 Specific examples of the charge-transporting material according to the present invention are given below, but are not necessarily limited thereto.

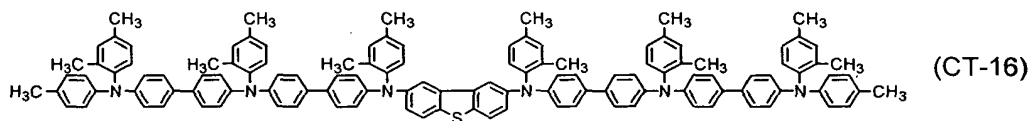
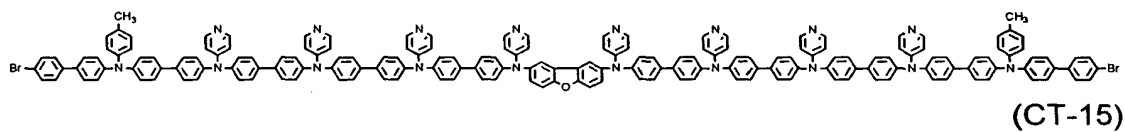
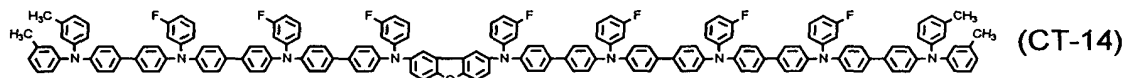
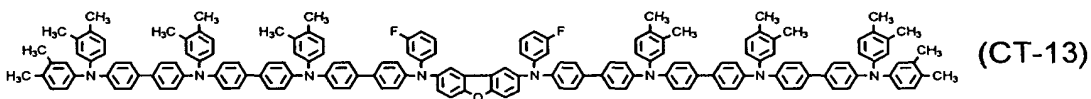
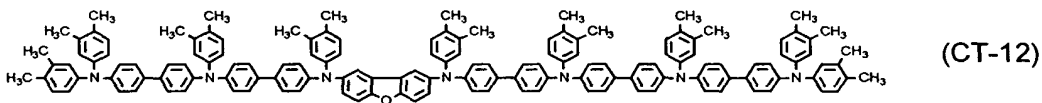
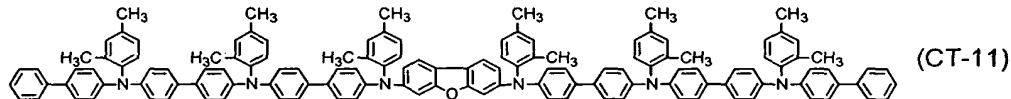


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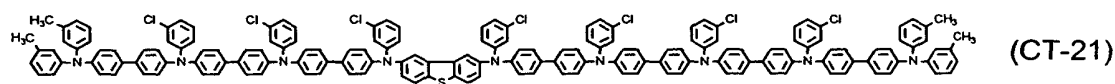
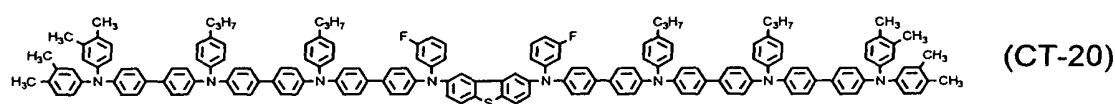
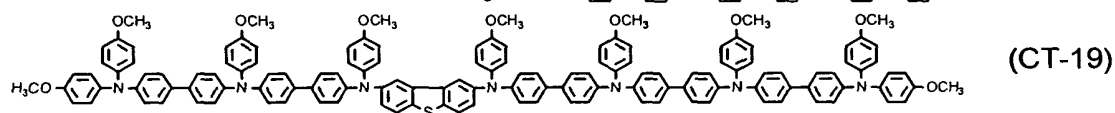
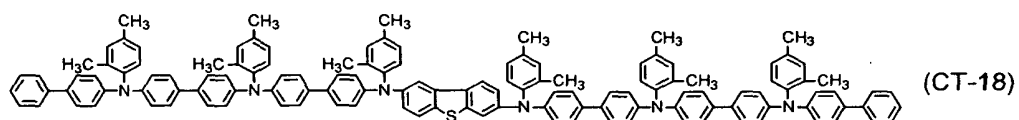
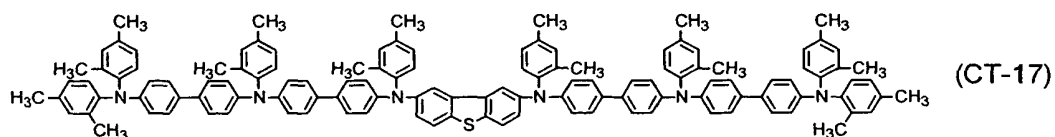




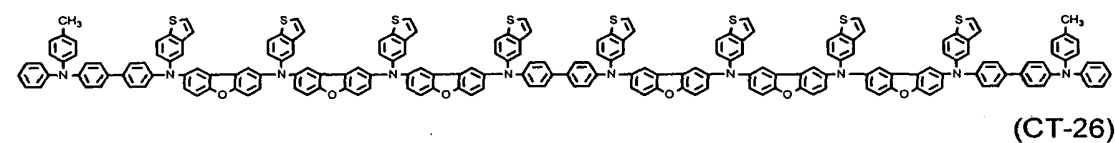
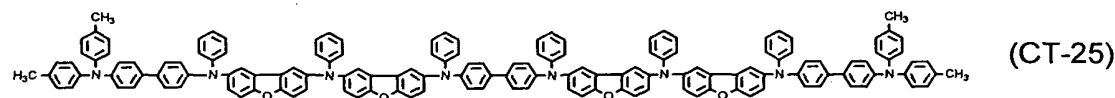
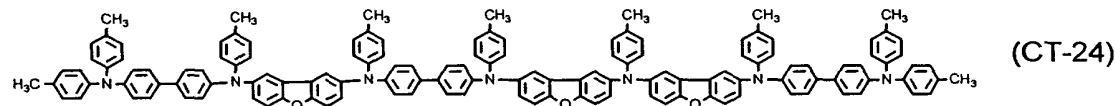
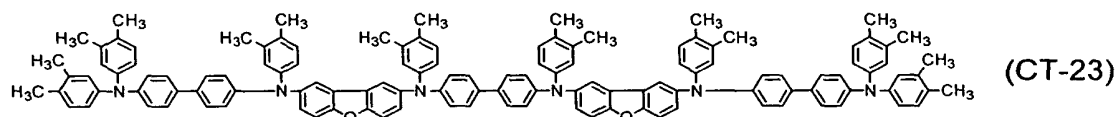
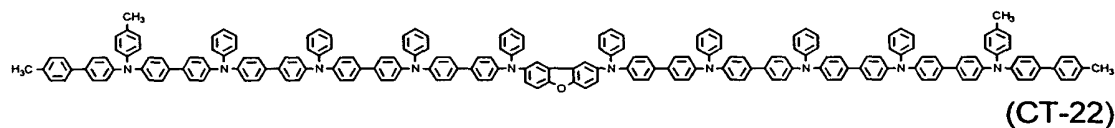
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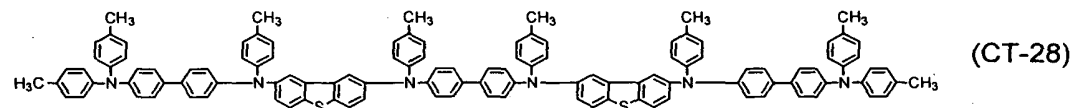
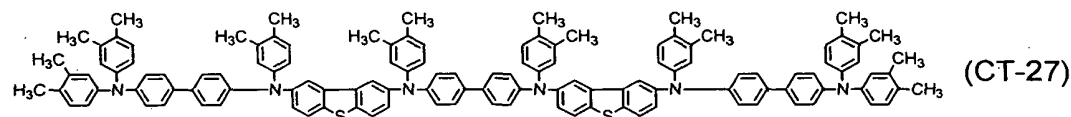
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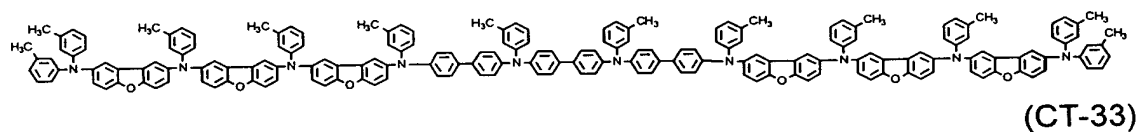
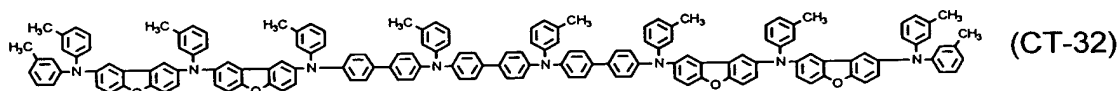
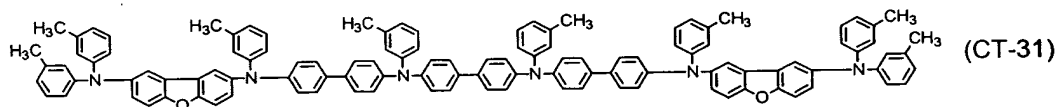
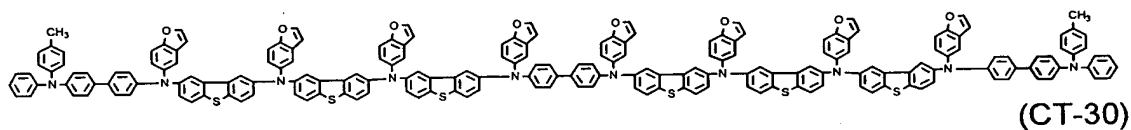
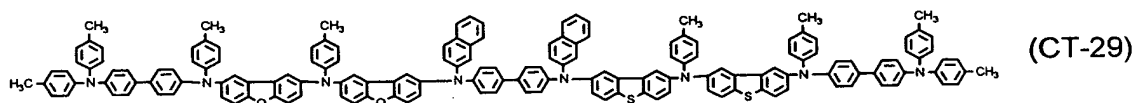


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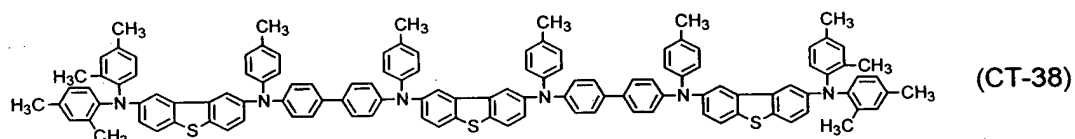
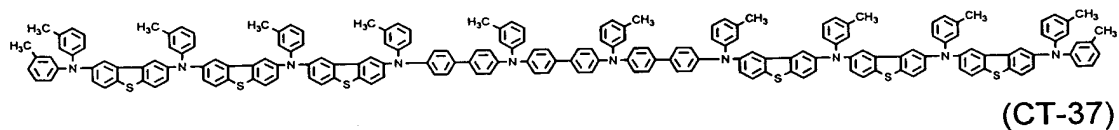
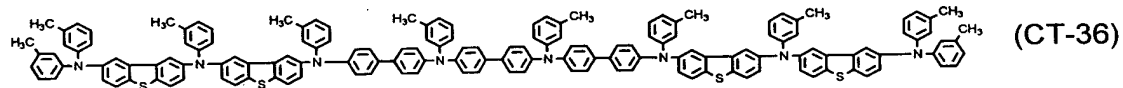
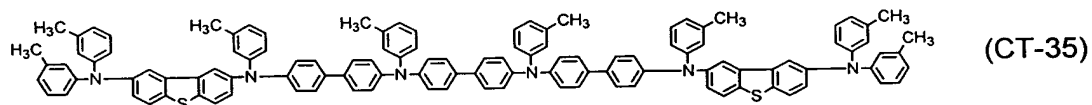
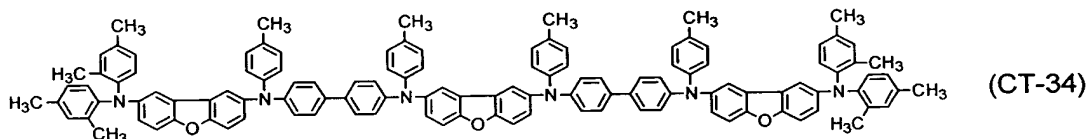


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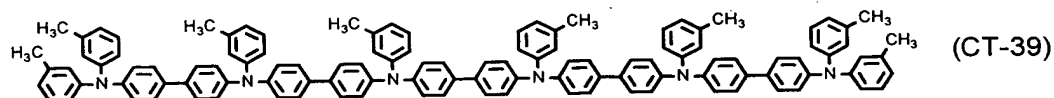


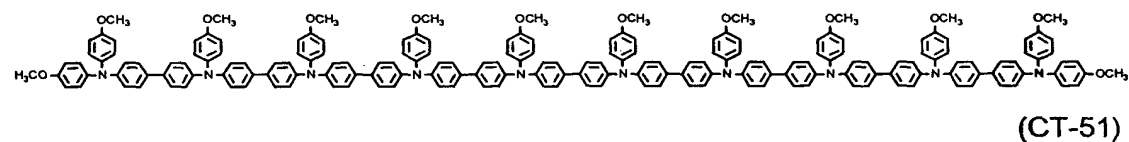
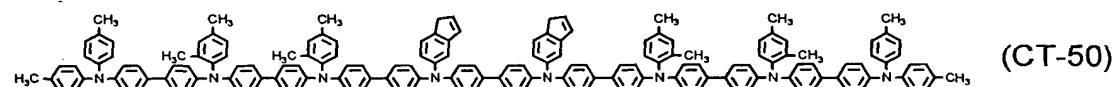
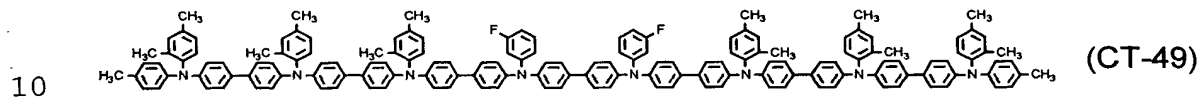
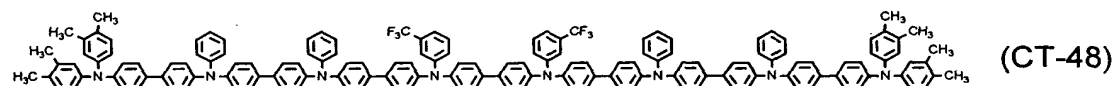
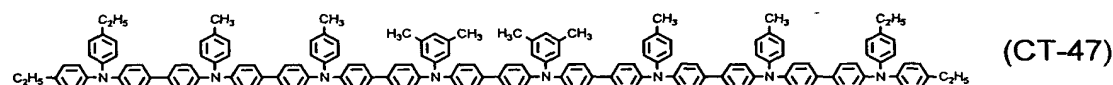
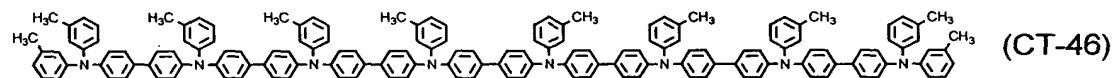
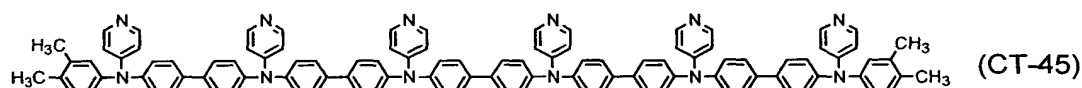
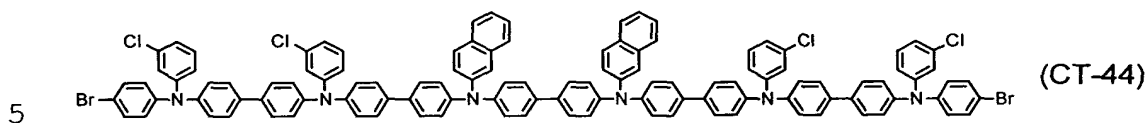
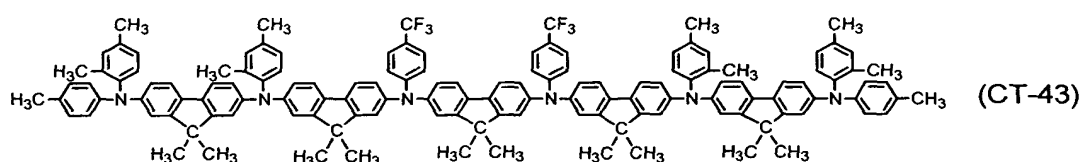
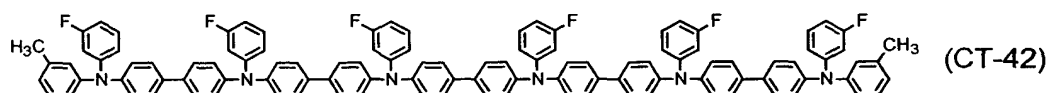
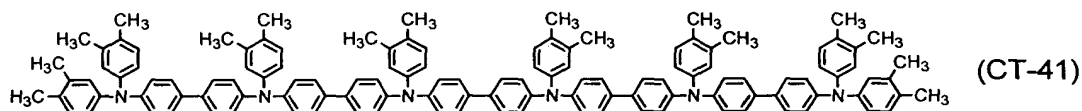
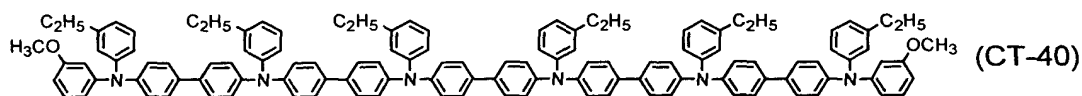


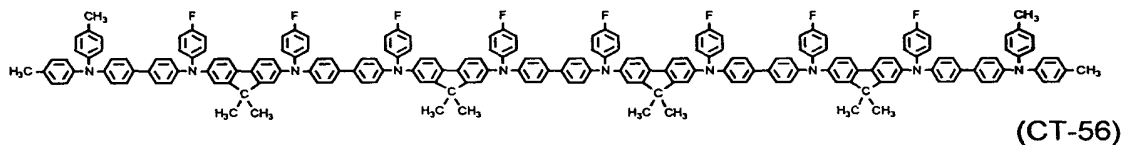
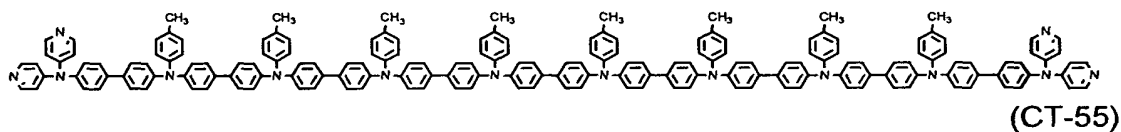
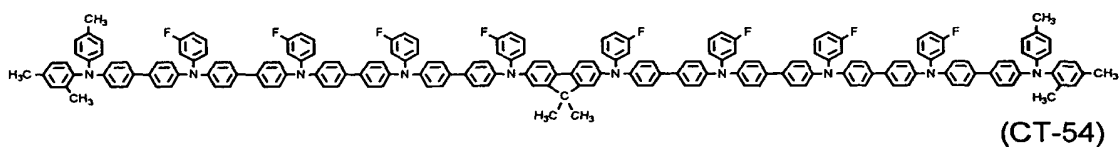
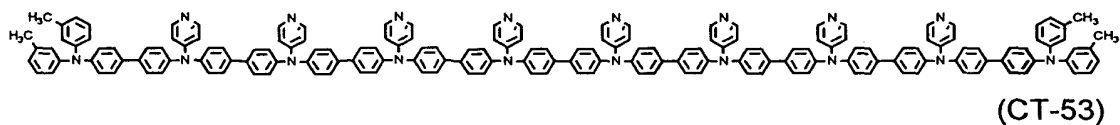
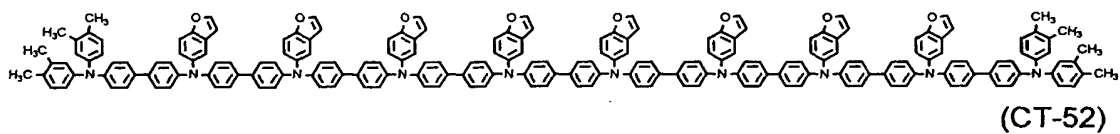
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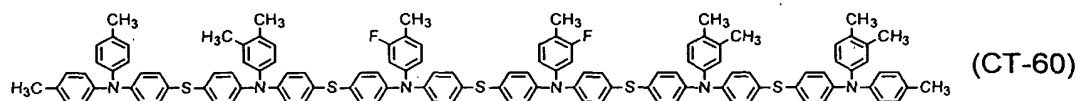
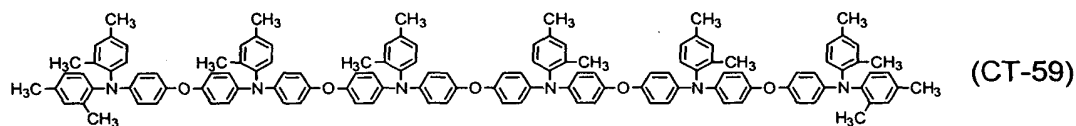
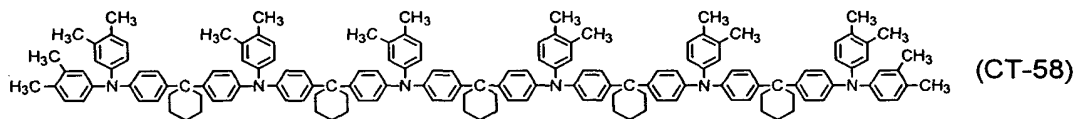
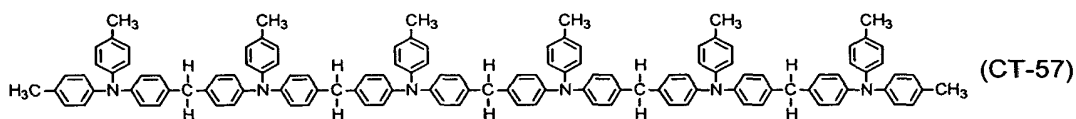
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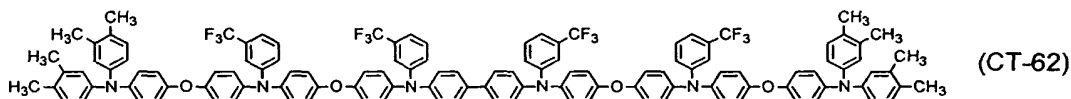
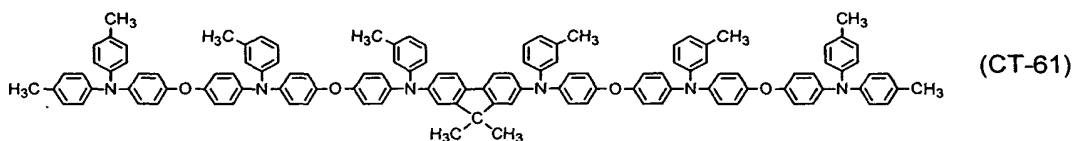






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The charge-transporting material in the present
 5 invention may preferably have a molecular weight of from
 1,500 to 4,000, and more preferably from 1,500 to 3,500.

The charge-transporting material in the present
 invention is characterized by being a
 high-molecular-weight charge-transporting material having
 10 a high singleness (being structurally single),
 represented by only a specific chemical structural
 formula. Hence, this charge-transporting material is
 difficult to produce by a production process carried out
 by repeated polymerization reaction of a monomer.
 15 Accordingly, it is preferable to use a
 charge-transporting material synthesized by a successive
 synthesis method which repeatedly carries out a synthesis
 process having conventionally been used in producing a
 low-molecular-weight charge-transporting material. The
 20 successive synthesis method is a method for synthesis in
 which the reaction of a raw material with a material to
 be reacted is carried out in a multi-stage process to
 form a single compound as a chief product. It differs

from a production method of synthesizing by
polymerization reaction a compound having molecular
weight distribution, and enables selective production of
a charge-transporting material having so high singleness
5 as not to have molecular weight distribution.

In the synthesis reaction used in the successive
synthesis method, synthesis reaction having
conventionally been used in producing
low-molecular-weight charge-transporting materials is
10 used. More specifically, Ullmann reaction or synthesis
using a metal catalyst is used. The multi-stage synthesis
may successively be continued to produce the intended
charge-transporting material, or, after one step of
synthesis has been completed, the step of purification
15 may be inserted before the step of next-stage synthesis.
Any purification methods conventionally commonly used may
also be used after the final step has been completed.
More specifically, there may be applied means such as
treatment with an adsorbent such as activated clay,
20 activated carbon, silica or alumina, purification by
column chromatography making use of silica or alumina or
gel permeation column chromatography making use of fine
polystyrene particles, and purification by
recrystallization or crystallization.

25 Production Examples of the charge-transporting
material in the present invention are shown below. The
present invention is by no means limited to these.

Production Example 1

Production of Exemplary Compound (CT-10) of
charge-transporting material:

- Production of bis(2,4-dimethylphenyl)amine -

5 Into a 1 liter three-necked reaction vessel fitted
with a cooling tube and a mechanical stirrer, 133 g (1.1
mols) of 2,4-dimethylphenylamine, 185 g (1.0 mol) of
1-bromo-2,4-dimethylbenzene, 11.2 g (0.05 mol) of
palladium acetate, 32.4 g (0.2 mol) of
10 2-tert-butylphosphinoyl-2-methylpropane, 212 g (1.0 mol)
of tripotassium phosphate and 500 mL of dimethylformamide
were introduced to effect reflux for 12 hours in a
nitrogen gas atmosphere and with heating in an oil bath.
After the reaction was completed, the reaction mixture
15 was left to cool to room temperature, followed by
extraction with toluene/water, then washing with
hydrochloric acid water, and thereafter distillation of
the organic layer under reduced pressure to give the
desired bis(2,4-dimethylphenyl)amine. The compound
20 obtained was in a yield of 189 g corresponding to 84%.
The compound obtained was measured with an elementary
analyzer (CHN CORDER MT-5, manufactured by Yanako K.K.).
In the following elementary analyses, the units of the
found values and calculated values are % by weight.
25 Found: C, 85.24; H, 8.53; N, 6.63
(Calculated: C, 85.28; H, 8.50; N, 6.22)

- Production of

bis(2,4-dimethylphenyl)[4-(bromophenyl)phenyl]amine -

Into a 2 liter three-necked reaction vessel fitted with a cooling tube and a mechanical stirrer, 112.5 g (0.5 mol) of bis(2,4-dimethylphenyl)amine, 156 g (0.5
5 mol) of 4,4-dibromophenyl, 5.6 g (0.025 mol) of palladium acetate, 27.7 g (0.05 mol) of bis(diphenylphosphino)ferrocene, 63.7 g (0.7 mol) of tert-butoxysodium (sodium tert-butoxide) and 800 mL of xylene were introduced to effect reflux for 5 hours in a
10 nitrogen gas atmosphere and with heating in an oil bath. After the reaction was completed, the reaction mixture was left to cool to room temperature, followed by extraction with toluene/water, then washing with hydrochloric acid water, and thereafter distillation of
15 the organic layer under reduced pressure and purification with a silica gel column (developer solvent: hexane/toluene = 2:1) to give the desired bis(2,4-dimethylphenyl)[4-(bromophenyl)phenyl]amine. The compound obtained was in a yield of 155.2 g corresponding
20 to 68%. The compound obtained was measured with the same elementary analyzer as the above.

Found: C, 73.44; H, 5.62; N, 2.98

(Calculated: C, 73.68; H, 5.74; N, 3.07)

- Production of

25 bis(2,4-dimethylphenyl)(4-{4-[(2,4-dimethylphenyl)amino]phenyl}phenyl)amine -

Into a 1 liter three-necked reaction vessel fitted

with a cooling tube and a mechanical stirrer, 91.3 g (0.2 mol) of bis(2,4-dimethylphenyl)[4-(bromophenyl)phenyl]amine, 36.3 g (0.3 mol) of 2,4-dimethylphenylamine, 2.24 g (0.01 mol) of palladium acetate, 11.1 g (0.02 mol) of bis(diphenylphosphino)ferrocene, 26.9 g (0.28 mol) of tert-butoxysodium and 500 mL of xylene were introduced to effect reflux for 3 hours in a nitrogen gas atmosphere and with heating in an oil bath. After the reaction was completed, the reaction mixture was left to cool to room temperature, followed by extraction with toluene/water, then washing with hydrochloric acid water, and thereafter distillation of the organic layer under reduced pressure and purification with a silica gel column (developer solvent: hexane/toluene = 1:1) to give the desired bis(2,4-dimethylphenyl)(4-{4-[(2,4-dimethylphenyl)amino]phenyl}phenyl)amine. The compound obtained was in a yield of 83.4 g corresponding to 84%. The compound obtained was measured with the same elementary analyzer as the above.

Found: C, 87.03; H, 7.26; N, 5.71
(Calculated: C, 87.05; H, 7.31; N, 5.64)

- Production of bis(2,4-dimethylphenyl)[4-(4-{(2,4-dimethylphenyl)[4-(4-bromophenyl)phenyl]amino}phenyl)phenyl]amine -

Into a 1 liter three-necked reaction vessel fitted with a cooling tube and a mechanical stirrer, 49.7 g (0.1 mol) of

bis(2,4-dimethylphenyl)(4-{4-[(2,4-dimethylphenyl)amino]phenyl}phenyl)amine, 31.2 g (0.1 mol) of 4,4'-dibromophenyl, 1.13 g (0.005 mol) of palladium acetate, 5.54 g (0.01 mol) of

5 bis(diphenylphosphino)ferrocene, 13.4 g (0.14 mol) of tert-butoxysodium and 300 mL of xylene were introduced to effect reflux for 5 hours in a nitrogen gas atmosphere and with heating in an oil bath. After the reaction was completed, the reaction mixture was left to cool to room

10 temperature, followed by extraction with toluene/water, then washing with hydrochloric acid water, and thereafter distillation of the organic layer under reduced pressure and purification with a silica gel column (developer solvent: hexane/toluene = 2:1) to give the desired

15 bis(2,4-dimethylphenyl)[4-(4-{(2,4-dimethylphenyl)[4-(4-bromophenyl)phenyl]amino}phenyl)phenyl]amine. The compound obtained was in a yield of 50.9 g corresponding to 70%. The compound obtained was measured with the same elementary analyzer as the above.

20 Found: C, 79.26; H, 5.98; N, 3.83
(Calculated: C, 79.22; H, 5.96; N, 3.85)

- Production of

(2,4-dimethylphenyl){8-[(2,4-dimethylphenyl)amino]dibenzo[b,d]furan-2-yl}amine -

25 Into a 500 mL three-necked reaction vessel fitted with a cooling tube and a mechanical stirrer, 65.2 g (0.2 mol) of 2,8-dibromodibenzofuran, 72.6 g (0.6 mol) of

2,4-dimethylphenylamine, 2.24 g (0.01 mol) of palladium acetate, 12.2 g (0.04 mol) of tri(o-tolyl)phosphine, 51.8 g (0.54 mol) of tert-butoxysodium and 300 mL of xylene were introduced to effect reflux for 5 hours in a nitrogen gas atmosphere and with heating in an oil bath. After the reaction was completed, an excess amount of 2,4-dimethylphenylamine and the solvent were distilled away under reduced pressure, and the reaction mixture was left to cool to room temperature, followed by extraction with toluene/water, then washing with hydrochloric acid water, and thereafter distillation of the solvent from the organic layer under reduced pressure and also recrystallization to give the desired (2,4-dimethylphenyl){8-[(2,4-dimethylphenyl)amino]dibenzo[b,d]furan-2-yl}amine. The compound obtained was in a yield of 71.5 g corresponding to 88%. The compound obtained was measured with the same elementary analyzer as the above.

Found: C, 82.68; H, 6.44; N, 6.92

(Calculated: C, 82.73; H, 6.45; N, 6.89)

- Production of Exemplary Compound (CT-10) of Charge-transporting Material -

Into a 500 mL three-necked reaction vessel fitted with a cooling tube and a mechanical stirrer, 36.4 g (0.05 mol) of bis(2,4-dimethylphenyl)[4-(4-[(2,4-dimethylphenyl)[4-(4-bromophenyl)phenyl]amino)phenyl]phenylamine, 10.2 g

(0.025 mol) of
(2,4-dimethylphenyl){8-[(2,4-dimethylphenyl)amino]dibenzo
[b,d]furan-2-yl}amine, 0.57 g (0.0025 mol) of palladium
acetate, 3.0 g (0.01 mol) of
5 biphenyl-2-yl-ditert-butylphosphine, 6.7 g (0.07 mol) of
tert-butoxysodium and 200 mL of xylene were introduced to
effect reflux for 4 hours in a nitrogen gas atmosphere
and with heating in an oil bath. After the reaction was
completed, the reaction mixture was left to cool to room
10 temperature, followed by extraction with toluene/water,
then washing with hydrochloric acid water, and thereafter
distillation of the solvent from the organic layer under
reduced pressure and purification with a silica gel
column (developer solvent: hexane/toluene = 1:1) to give
15 the desired charge-transporting material shown as the
exemplary compound (CT-10). The compound obtained was in
a yield of 39.1 g corresponding to 89%. The compound
obtained was measured with the same elementary analyzer
as the above.
20 Found: C, 84.80; H, 5.81; N, 4.83
(Calculated: C, 84.85; H, 5.85; N, 4.79)

Mass analysis of the compound obtained was made
using a laser desorption ionization time-of-flight mass
(MALDI-TOF-MASS) spectrometer (REFLEX III; manufactured
25 by Bruker Co.; matrix: 9-nitroanthracene). The mass
spectrum measurement results obtained are shown in Fig. 1
where the abscissa indicates the mass-to-charge ratio

[mass (m) of ions formed/valence (z) of ions formed] and the ordinate indicates the intensity of ions formed and detected.

Production Example 2

- 5 Production of Exemplary Compound (CT-17) of charge-transporting material:

- Production of

(2,4-dimethylphenyl){8-[(2,4-dimethylphenyl)amino]dibenzo
[b]benzo[b]thiophen-2-yl}amine -

- 10 Into a 500 mL three-necked reaction vessel fitted with a cooling tube and a mechanical stirrer, 68.4 g (0.2 mol) of 2,8-dibromodibenzothiophene, 72.6 g (0.6 mol) of 2,4-dimethylphenylamine, 2.24 g (0.01 mol) of palladium acetate, 12.2 g (0.04 mol) of tri(o-tolyl)phosphine, 51.8
15 g (0.54 mol) of tert-butoxysodium and 300 mL of xylene were introduced to effect reflux for 5 hours in a nitrogen gas atmosphere and with heating in an oil bath. After the reaction was completed, an excess amount of 2,4-dimethylphenylamine and the solvent were distilled
20 away under reduced pressure, and the reaction mixture was left to cool to room temperature, followed by extraction with toluene/water, then washing with hydrochloric acid water, and thereafter distillation of the solvent from the organic layer under reduced pressure and also
25 recrystallization to give the desired
(2,4-dimethylphenyl){8-[(2,4-dimethylphenyl)amino]dibenzo
[b]benzo[b]thiophen-2-yl}amine. The compound obtained was

in a yield of 71.0 g corresponding to 84%. The compound obtained was measured with the same elementary analyzer as the above.

Found: C, 79.60; H, 6.24; N, 6.52

5 (Calculated: C, 79.58; H, 6.20; N, 6.63)

- Production of Exemplary Compound (CT-17) of Charge-transporting Material -

Into a 500 mL three-necked reaction vessel fitted with a cooling tube and a mechanical stirrer, 36.4 g
10 (0.05 mol) of bis(2,4-dimethylphenyl)[4-(4-((2,4-dimethylphenyl)[4-(4-bromophenyl)phenyl]amino)phenyl)phenyl]amine, 10.6 g (0.025 mol) of (2,4-dimethylphenyl){8-[(2,4-dimethylphenyl)amino]dibenzo
15 [b]benzo[b]thiophen-2-yl}amine, 0.57 g (0.0025 mol) of palladium acetate, 3.0 g (0.01 mol) of biphenyl-2-yl-ditert-butylphosphine, 6.7 g (0.07 mol) of tert-butoxysodium and 200 mL of xylene were introduced to effect reflux for 4 hours in a nitrogen gas atmosphere
20 and with heating in an oil bath. After the reaction was completed, the reaction mixture was left to cool to room temperature, followed by extraction with toluene/water, then washing with hydrochloric acid water, and thereafter distillation of the solvent from the organic layer under
25 reduced pressure and purification with a silica gel column (developer solvent: hexane/toluene = 1:1) to give the desired charge-transporting material shown as the

exemplary compound (CT-17). The compound obtained was in a yield of 37.3 g corresponding to 87%. The compound obtained was measured with the same elementary analyzer as the above.

- 5 Found: C, 86.80; H, 6.40; N, 4.86
(Calculated: C, 86.78; H, 6.46; N, 4.90)

Mass analysis of the compound obtained was also made using the same spectrometer as the above. The mass spectrum measurement results obtained are shown in Fig. 2.

10 Production Example 3

Production of Exemplary Compound (CT-39) of charge-transporting material:

- Production of (2,4-dimethylphenyl)-p-tolylamine -

- Into a 1 liter three-necked reaction vessel fitted
15 with a cooling tube and a mechanical stirrer, 133 g (1.1 mols) of 2,4-dimethylphenylamine, 171 g (1.0 mol) of 1-bromo-4-methylbenzene, 11.2 g (0.05 mol) of palladium acetate, 32.4 g (0.2 mol) of
2-tert-butylphosphinoyl-2-methylpropane, 212 g (1.0 mol)
20 of tripotassium phosphate and 500 mL of dimethylformamide were introduced to effect reflux for 12 hours in a nitrogen gas atmosphere and with heating in an oil bath. After the reaction was completed, the reaction mixture was left to cool to room temperature, followed by
25 extraction with toluene/water, then washing with hydrochloric acid water, and thereafter distillation of the organic layer under reduced pressure to obtain the

desired (2,4-dimethylphenyl)-p-tolylamine. The compound obtained was in a yield of 189.5 g corresponding to 84%. The compound obtained was measured with the same elementary analyzer as the above.

5 Found: C, 85.27; H, 8.09; N, 6.64

(Calculated: C, 85.26; H, 8.11; N, 6.63)

- Production of

(4'-(bromobiphenyl-4-yl)-(2,4-dimethylphenyl)-p-tolylamine -

10 Into a 2 liter three-necked reaction vessel fitted with an air-cooling tube and a mechanical stirrer, 105.7 g (0.5 mol) of (2,4-dimethylphenyl)-p-tolylamine, 215.4 g (0.6 mol) of 4'-bromo-4-iodobiphenyl, 50 g of copper powder, 60 g of potassium carbonate and 600 mL of
15 o-dichlorobenzene were introduced and heated for 12 hours in an oil bath. After the reaction was completed, the reaction mixture was left to cool to room temperature, followed by removing solid matter by filtration, and thereafter extraction with toluene/water. Then, the
20 solvent was distilled away from the organic layer under reduced pressure, followed by purification with a silica gel column (developer solvent: hexane/toluene = 2:1) to give the desired
(4'-(bromobiphenyl-4-yl)-(2,4-dimethylphenyl)-p-tolylamine
25 e. The compound obtained was in a yield of 165.9 and corresponding to 75%. The compound obtained was measured with the same elementary analyzer as the above.

Found: C, 73.33; H, 5.48; N, 3.17

(Calculated: C, 73.30; H, 5.47; N, 3.17)

- Production of

$N^4, N^{4'}$ -bis(2,4-dimethylphenyl)- N^4 -p-tolylbiphenyl-4,4'-diam
5 ine -

Into a 1 liter three-necked reaction vessel fitted
with a cooling tube and a mechanical stirrer, 88.5 g (0.2
mol) of

(4'-(bromobiphenyl-4-yl)-(2,4-dimethylphenyl)-p-tolylamin
10 e, 26.7 g (0.22 mol) of 2,4-dimethylphenylamine, 2.2 g
(0.01 mol) of palladium acetate, 11.1 g (0.02 mol) of
bis(diphenylphosphino)ferrocene, 26.8 g (0.28 mol) of
tert-butoxysodium and 400 mL of xylene were introduced to
effect reflux for 5 hours in a nitrogen gas atmosphere
15 and with heating in an oil bath. After the reaction was
completed, the reaction mixture was left to cool to room
temperature, followed by extraction with toluene/water,
then washing with hydrochloric acid water, and thereafter
distillation of the organic layer under reduced pressure
20 and purification with a silica gel column (developer
solvent: hexane/toluene = 1:1) to give the desired
 $N^4, N^{4'}$ -bis(2,4-dimethylphenyl)- N^4 -p-tolylbiphenyl-4,4'-diam
ine. The compound obtained was in a yield of 71.4 g
corresponding to 74%. The compound obtained was measured
25 with the same elementary analyzer as the above.

Found: C, 87.05; H, 7.14; N, 5.81

(Calculated: C, 87.10; H, 7.10; N, 5.80)

- Production of

$N^{4'}-(4'\text{-bromobiphenyl-4-yl})-N^4,N^{4'}\text{-bis}(2,4\text{-dimethylphenyl})-N^4\text{-p-tolylbiphenyl-4,4'-diamine}$ -

Into a 2 liter three-necked reaction vessel fitted
5 with an air-cooling tube and a mechanical stirrer, 48.3 g
(0.1 mol) of
 $N^4,N^{4'}\text{-bis}(2,4\text{-dimethylphenyl})-N^4\text{-p-tolylbiphenyl-4,4'-diamine}$, 39.5 g (0.11 mol) of 4'-bromo-4-iodobiphenyl, 20 g
of copper powder and 200 mL of o-dichlorobenzene were
10 introduced and heated for 12 hours in an oil bath. After
the reaction was completed, the reaction mixture was left
to cool to room temperature, followed by removing solid
matter by filtration, and thereafter extraction with
toluene/water. Then, the solvent was distilled away from
15 the organic layer under reduced pressure, followed by
purification with a silica gel column (developer solvent:
hexane/toluene = 2:1) to give the desired
 $N^{4'}-(4'\text{-bromobiphenyl-4-yl})-N^4,N^{4'}\text{-bis}(2,4\text{-dimethylphenyl})-N^4\text{-p-tolylbiphenyl-4,4'-diamine}$. The compound obtained
20 was in a yield of 45.0 g corresponding to 63%. The
compound obtained was measured with the same elementary
analyzer as the above.

Found: C, 78.88; H, 5.69; N, 3.96

(Calculated: C, 79.09; H, 5.79; N, 3.92)

25 - Production of

$N^4,N^{4'}\text{-bis}(2,4\text{-dimethylphenyl})\text{-biphenyl-4,4'-diamine}$ -

Into a 1 liter three-necked reaction vessel fitted

with a cooling tube and a mechanical stirrer, 62.4 g (0.2 mol) of 4,4'-dibromobiphenyl, 72.6 g (0.6 mol) of 2,4-dimethylphenylamine, 2.24 g (0.01 mol) of palladium acetate, 12.2 g (0.04 mol) of tri(o-tolyl)phosphine, 51.8 g (0.54 mol) of tert-butoxysodium and 300 mL of xylene were introduced to effect reflux for 5 hours in a nitrogen gas atmosphere and with heating in an oil bath. After the reaction was completed, an excess amount of 2,4-dimethylphenylamine and the solvent were distilled away under reduced pressure, and the reaction mixture was left to cool to room temperature, followed by extraction with toluene/water, then washing with hydrochloric acid water, and thereafter distillation of the solvent from the organic layer under reduced pressure and also recrystallization to give the desired N⁴,N^{4'}-bis(2,4-dimethylphenyl)-biphenyl-4,4'-diamine. The compound obtained was in a yield of 62.7 g corresponding to 80%. The compound obtained was measured with the same elementary analyzer as the above.

Found: C, 85.66; H, 7.18; N, 7.16

(Calculated: C, 85.67; H, 7.19; N, 7.14)

- Production of Exemplary Compound (CT-39) of Charge-transporting Material -

Into a 500 mL three-necked reaction vessel fitted with a cooling tube and a mechanical stirrer, 35.7 g (0.05 mol) of N⁴-(4'-bromobiphenyl-4-yl)-N⁴,N^{4'}-bis(2,4-dimethylphenyl)-

N⁴-p-tolylbiphenyl-4,4'-diamine, 9.8 g (0.025 mol) of
N⁴,N⁴'-bis(2,4-dimethylphenyl)-biphenyl-4,4'-diamine, 0.57
g (0.0025 mol) of palladium acetate, 3.0 g (0.01 mol) of
biphenyl-2-yl-ditert-butylphosphine, 6.7 g (0.07 mol) of
5 tert-butoxysodium and 200 mL of xylene were introduced to
effect reflux for 4 hours in a nitrogen gas atmosphere
and with heating in an oil bath. After the reaction was
completed, the reaction mixture was left to cool to room
temperature, followed by extraction with toluene/water,
10 then washing with hydrochloric acid water, and thereafter
distillation of the solvent from the organic layer under
reduced pressure and purification with a silica gel
column (developer solvent: hexane/toluene = 1:1) to give
the desired charge-transporting material shown as the
15 exemplary compound (CT-39). The compound obtained was in
a yield of 34.4 g corresponding to 83%. The compound
obtained was measured with the same elementary analyzer
as the above.

Found: C, 88.40; H, 6.51; N, 5.09

20 (Calculated: C, 88.37; H, 6.56; N, 5.07)

Mass analysis of the compound obtained was made
using the same spectrometer as the above. The mass
spectrum measurement results obtained are shown in Fig. 3.

In the present invention, in respect of the
25 improvement of running (extensive operation) lifetime of
the electrophotographic photosensitive member, in
particular, remarkable prevention of image quality from

lowering because of scratches made on the electrophotographic photosensitive member surface, in respect of the achievement of higher sensitivity of the electrophotographic photosensitive member, and in respect
5 of the superior image stability, these are considered in the following way.

The high charge transport performance of the charge-transporting material that contributes to the achievement of higher sensitivity of the
10 electrophotographic photosensitive member is brought out by the tertiary amine in which either the aromatic carbocyclic groups or the aromatic heterocyclic groups are bonded to the nitrogen atom, and the charge transport performance can be made highly efficient by arranging
15 such tertiary amines in the molecule in a large number. However, arranging the tertiary amines in the molecule in a large number tends to lower the solubility of the charge-transporting material in the solvent. This may also lower the compatibility of the charge-transporting
20 material with the binder resin after a coating solution or dispersion for forming a photosensitive layer of the electrophotographic photosensitive member has been applied on its support, tending to form a separate state, and to make poor the performance of the resulting
25 electrophotographic photosensitive member. In the present invention, the charge-transporting material has the tertiary amine in its molecule in an appropriate number,

and hence the electrophotographic photosensitive member can be made to have higher sensitivity in virtue of its charge transport performance having been made higher efficient, without lowering the solubility of the
5 charge-transporting material in the solvent or lowering its compatibility with the binder resin.

In addition, in respect of the remarkably improved prevention of image quality from lowering because of scratches made on the electrophotographic photosensitive
10 member surface, this is considered to be due to the rigid molecular structure of the charge-transporting material in the present invention, coming from the structure represented by any of the above Formulas (1) to (5), and resulting in the improvement of running lifetime that has
15 not been achievable by conventionally used electrophotographic photosensitive members in which low-molecular-weight charge-transporting material has been dispersed in the binder resin.

It is also considered that the present invention
20 has such features that the charge-transporting material has a molecular weight of from 1,500 to 4,000 and the charge-transporting material represented by any of the above Formulas (1) to (5) is held in a proportion of from 90% by weight to 100% by weight based on the total weight
25 of the charge-transporting material(s) contained in the photosensitive layer, thereby achieving the above improvement in performance. This is considered to be for

the reason that if the proportion held by any charge-transporting material other than the charge-transporting material represented by any of the above Formulas (1) to (5) is 10% by weight or more based
5 on the total weight of the charge-transporting material(s) contained in the photosensitive layer, the properties of the charge-transporting material according to the present invention, which contribute to the achievement of higher sensitivity, may be inhibited. For
10 the same reason, it is considered that at the same time the charge-transporting material used in the present invention has superiority also in regard to image stability, in particular, image stability in a high humidity environment. In regard to the improvement of
15 running lifetime of the electrophotographic photosensitive member, it is also considered that the presence of any charge-transporting materials other than the charge-transporting material represented by any of the above Formulas (1) to (5) brings about a low
20 mechanical strength.

The electrophotographic photosensitive member of the present invention is constructed as described below.

The photosensitive layer of the present invention may be in the form of either a single-layer type in which
25 a charge-generating material and a charge-transporting material are contained in the same layer, or a function-separated type (multi-layer type), which is

functionally separated into a charge generation layer containing a charge-generating material and a charge transport layer containing a charge-transporting material. The function-separated type (multi-layer type) is

5 preferred in view of electrophotographic performance.

More preferred is a function-separated type in which the charge generation layer and the charge transport layer are formed in this order from the support side. In the following, when expressed as the function-separated type
10 (multi-layer type), it means that the charge generation layer and the charge transport layer are formed in this order from the support side.

The support used in the electrophotographic photosensitive member of the present invention may be any
15 materials having a conductivity, including, e.g., those obtained by molding metals such as aluminum, copper, chromium, nickel, zinc and stainless steel into drums or sheets, those obtained by laminating metal foil of aluminum or copper to plastic films, and those obtained
20 by vacuum-depositing aluminum, indium oxide or tin oxide on plastic films.

Where images are inputted through laser light as in the case of LBP (laser beam printers), a conductive layer may be provided on the support for the purpose of
25 preventing interference fringes due to light scattering or for the purpose of covering any scratches of the support. This layer may be formed of a binder resin in

which conductive particles such as carbon black and metal particles have been dispersed.

The conductive layer may preferably have a layer thickness of from 5 μm to 40 μm , and more preferably from 10 μm to 30 μm .

On the support or conductive layer, an intermediate layer having the function of bonding may also be provided. As materials for the intermediate layer, usable are polyamide, polyvinyl alcohol, polyethylene oxide, ethyl cellulose, casein, polyurethane and polyether-urethane. Any of these may be dissolved in a suitable solvent, and the resulting solution may be applied on the support or conductive layer, followed by drying to form the intermediate layer.

The intermediate layer may preferably have a layer thickness of from 0.05 μm to 5 μm , and more preferably from 0.3 μm to 1 μm .

In the case of the function-separated type (multi-layer type) photosensitive layer, the photosensitive layer is provided on the support, conductive layer or intermediate layer.

To form the charge generation layer, the charge-generating material may be sufficiently dispersed together with a binder resin of 0.3 to 4 times the amount of that charge-generating material and a solvent by means of a homogenizer, an ultrasonic dispersion machine, a ball mill, a vibrating ball mill, a sand mill, an

attritor, a roll mill, a liquid impact type high-speed dispersion machine or the like, and the dispersion obtained may be applied, followed by drying.

The charge generation layer may preferably have a
5 layer thickness of 5 μm or less, and more preferably from 0.1 μm to 2 μm .

As the charge-generating material, those commonly known may be used, which may include, e.g., pigments such as selenium-tellurium, pyrylium, a metal phthalocyanine,
10 metal-free phthalocyanine, anthanthrone, dibenzpirenequinone, trisazo, cyanine, disazo, monoazo, indigo and quinacridone pigments.

Any of these pigments may be sufficiently dispersed together with a binder resin of 0.3 to 4 times the amount
15 of that pigment and a solvent by means of a homogenizer, an ultrasonic dispersion machine, a ball mill, a vibrating mill, a sand mill, an attritor, a roll mill, a liquid impact type high-speed dispersion machine or the like to prepare a dispersion. In the case of the
20 function-separated type (multi-layer type) photosensitive layer, this dispersion may be applied on the support, conductive layer or intermediate layer, followed by drying to form the charge generation layer.

In the case of the function-separated type
25 (multi-layer type) photosensitive layer, the charge transport layer is formed on the charge generation layer.

The charge transport layer or photosensitive layer

does not necessarily require any binder resin when the charge transport layer itself has binding properties. However, from the viewpoint of mechanical strength and electrophotographic performance, it is preferable for the photosensitive layer to contain a binder resin. The binder resin may also preferably be insulating.

The photosensitive layer may also preferably be a surface layer of the electrophotographic photosensitive member.

To form the charge transport layer, the charge-transporting material described above and the binder resin may be dissolved in a solvent to prepare a coating solution, and this coating solution may be formed on the charge generation layer, followed by drying.

In the coating solution, the charge-transporting material and the binder resin may preferably be in a proportion (charge-transporting material/binder resin) of from 1/10 to 12/10 in weight ratio. From the viewpoint of charge transport performance of the electrophotographic photosensitive member or strength of the charge transport layer, it may more preferably be 2/10 to 10/10.

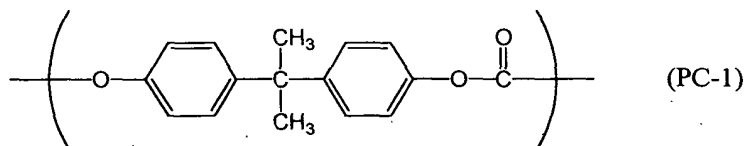
As the binder resin, any resins may be used as long as they are resins usually usable in photosensitive layers or charge transport layers, such as polycarbonate resins, polyarylate resins, polyester resins, polystyrene resins, polymethacrylate resins and polyacrylate resins. From the viewpoint of light transmission properties and

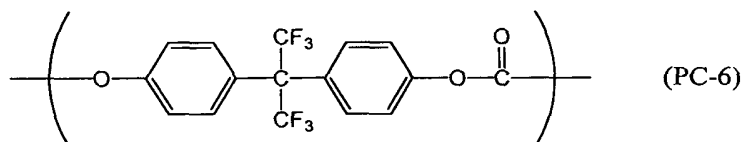
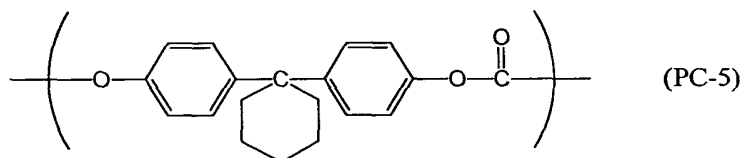
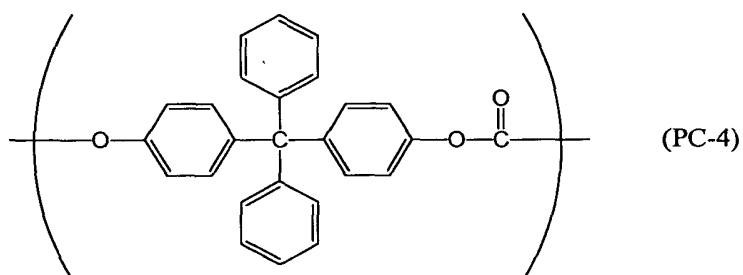
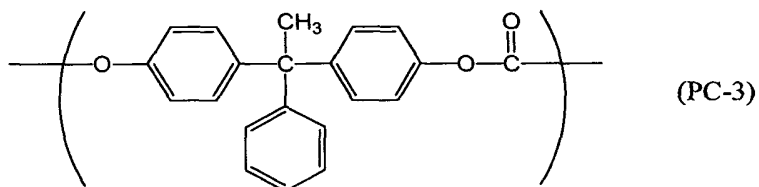
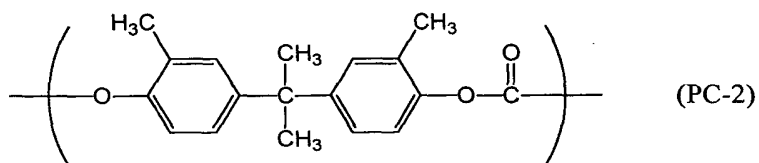
film-forming properties of the resin, and where the photosensitive layer or the charge transport layer is the surface layer of the electrophotographic photosensitive member, also from the viewpoint of wear resistance,
5 polycarbonate resins or polyarylate resins are preferred.

The polycarbonate resin may preferably have a viscosity-average molecular weight (Mv) of from 20,000 to 80,000.

The divalent organic residual group moiety included
10 in the polycarbonate resin may have any structure as long as it is a divalent organic residual group such as a substituted or unsubstituted divalent biphenyl residual group, a substituted or unsubstituted divalent bisphenyl residual group, a substituted or unsubstituted divalent
15 biphenyl ether residual group, or a substituted or unsubstituted divalent biphenyl thioether residual group. It may preferably be a substituted or unsubstituted divalent biphenyl residual group, a substituted or unsubstituted divalent bisphenyl residual group or a
20 substituted or unsubstituted divalent biphenyl ether residual group.

Specific examples of the repeating structural unit of the polycarbonate resin are shown below. Examples are by no means limited to these structures.





5

In order to improve productivity, a copolymer resin may also be used in which a divalent organic residual group of a different type is used in the divalent organic residual group moiety of the polycarbonate resin. In order to efficiently bring out the effect of blending, the blending proportion may preferably be from 5/95 to 95/5, and more preferably from 20/80 to 80/20.

The polyarylate resin may preferably have a

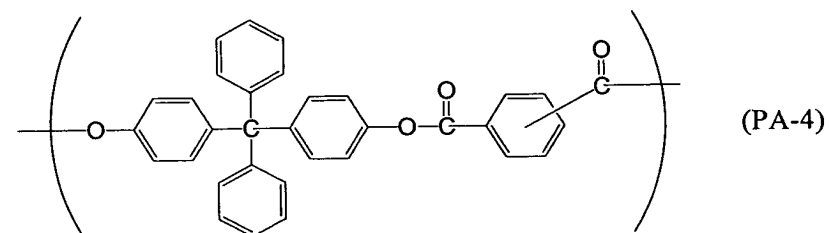
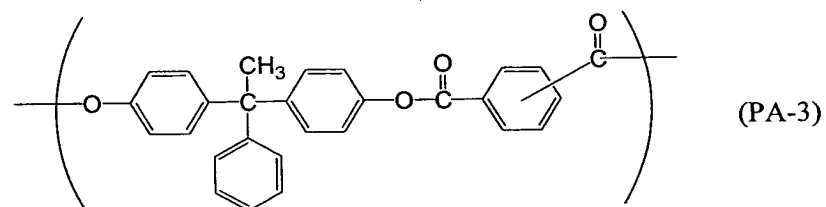
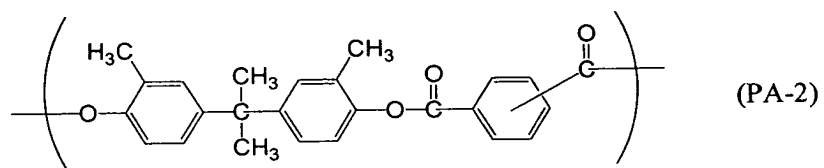
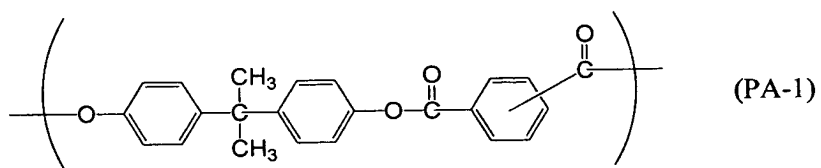
weight-average molecular weight (Mw) of from 50,000 to 200,000, and from the viewpoint of strength, productivity and so forth, more preferably from 80,000 to 150,000.

In regard to the structure of a phthalic-acid moiety used in the polyarylate resin, isophthalic acid and/or terephthalic acid may be used. The isophthalic acid and terephthalic acid in the resin may be in a proportion (isophthalic acid/terephthalic acid) of from 0/100 to 100/0 in weight ratio. From the viewpoint of strength of the polyarylate resin, the proportion of isophthalic acid/terephthalic acid may more preferably be from 30/70 to 70/30.

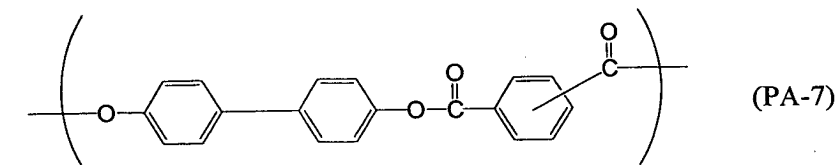
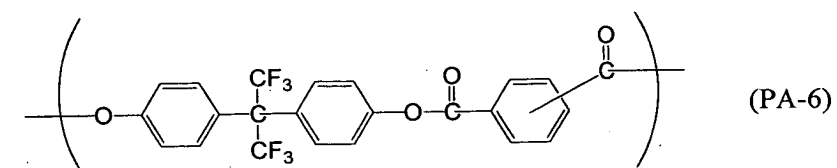
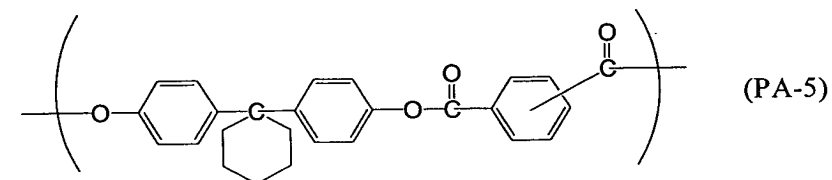
The divalent organic residual group moiety included in the polyarylate resin may have any structure as long as it is a divalent organic residual group such as a substituted or unsubstituted divalent biphenyl residual group, a substituted or unsubstituted divalent bisphenyl residual group, a substituted or unsubstituted divalent biphenyl ether residual group, or a substituted or unsubstituted divalent biphenyl thioether residual group. It may preferably be a substituted or unsubstituted divalent biphenyl residual group, a substituted or unsubstituted divalent bisphenyl residual group or a substituted or unsubstituted divalent biphenyl ether residual group.

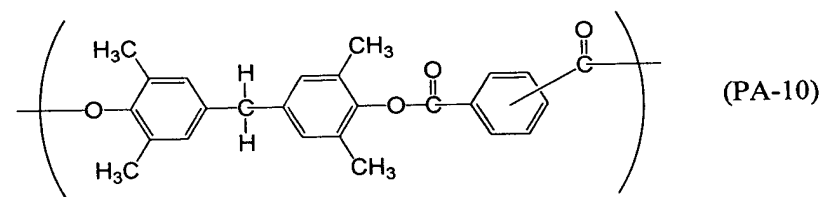
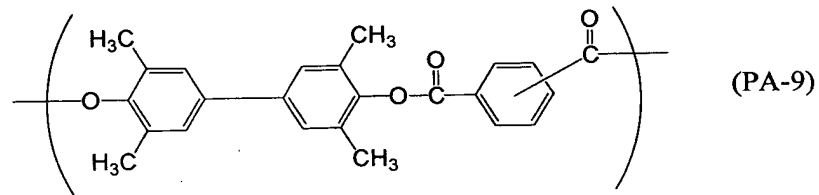
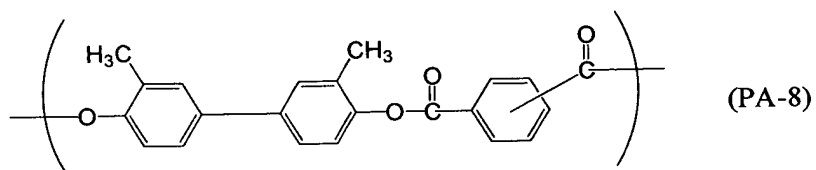
Specific examples of the repeating structural unit of the polyarylate resin are shown below. Examples are by

no means limited to these structures.



5





In order to improve solubility, a copolymer resin
 5 may also be used in which a divalent organic residual
 group of a different type is used in the divalent organic
 residual group moiety in the polyarylate resin. In order
 to efficiently bring out the effect of blending, the
 blending proportion may preferably be from 5/95 to 95/5,
 10 and more preferably from 20/80 to 80/20.

In addition, in order to improve productivity, a
 polyarylate resin or polycarbonate resin with a different
 structure may be blended in the above polyarylate resin
 or polycarbonate resin. In order to efficiently bring out
 15 the effect of blending, the blending proportion may
 preferably be from 5/95 to 95/5, and more preferably from
 20/80 to 80/20.

In the charge transport layer, an antioxidant, a
 heat stabilizer, an ultraviolet absorber and a

plasticizer may optionally be incorporated.

In the case when the charge transport layer is the surface layer of the electrophotographic photosensitive member, a lubricant or fine particles may optionally be
5 used. Such a lubricant or fine particles may include fine resin particles such as fine polytetrafluoroethylene particles and fine polystyrene particles, fine metal oxide particles such as fine silica particles, fine alumina particles and fine tin oxide particles, fine
10 particles obtained by subjecting any of these fine particles to surface treatment, solid lubricants such as zinc stearate, silicones substituted with an alkyl group, aliphatic oils having a fluorinated alkyl group, and varnishes.

15 The charge transport layer may preferably have a layer thickness of from 5 μm to 40 μm , and more preferably from 15 μm to 30 μm .

A layer for protecting the photosensitive layer, i.e., a protective layer may additionally be provided on
20 the photosensitive layer as a surface layer of the electrophotographic photosensitive member.

As resins used in the protective layer, thermoplastic resins or thermosetting resins and photo-curable resins are preferred. Further, more
25 preferred are polycarbonate resin, polyarylate resin, phenolic resins, acrylic resins and epoxy resins. For the purpose of reducing residual potential or improving film

strength, the protective layer may also be incorporated with conductive particles or a lubricant.

In forming the protective layer, a wet coating may be cured by heat, light or electron rays, and may
5 optionally be incorporated with a polymerization initiator and an antioxidant.

The solvent used in the steps of forming the respective layers of the electrophotographic photosensitive member may include chlorobenzene,
10 tetrahydrofuran, 1,4-dioxane, toluene and xylene, any of which may be used alone or a plurality of which may be used in combination.

As methods for applying the above layers, conventionally know methods may be used, such as dip
15 coating, spray coating and bar coating.

An electrophotographic apparatus having a process cartridge having the electrophotographic photosensitive member of the present invention is described below.

Fig. 4 schematically illustrates the construction
20 of such an electrophotographic apparatus having a process cartridge having the electrophotographic photosensitive member of the present invention.

In Fig. 4, reference numeral 1 denotes a drum-shaped electrophotographic photosensitive member of
25 the present invention, which is rotatively driven around an axis 2 in the direction of an arrow at a stated peripheral speed. The electrophotographic photosensitive

member 1 is, in the course of its rotation, uniformly electrostatically charged on its periphery to a positive or negative, given potential through a primary charging means 3. The electrophotographic photosensitive member
5 thus charged is then exposed to imagewise exposure light 4 emitted from an exposure means (not shown) for slit exposure or laser beam scanning exposure. In this way, electrostatic latent images are successively formed on the periphery of the electrophotographic photosensitive
10 member 1.

The electrostatic latent images thus formed are subsequently developed with toner by the operation of a developing means 5. The toner images thus formed by development are then successively transferred by the
15 operation of a transfer means 6, to a transfer material 7 fed from a paper feed section (not shown) to the part between the electrophotographic photosensitive member 1 and the transfer means 6 in such a manner as synchronized with the rotation of the electrophotographic
20 photosensitive member 1.

The transfer material 7 onto which the toner images have been transferred is separated from the surface of the electrophotographic photosensitive member, is led through an image fixing means 8, where the toner images
25 are fixed, and is then put out of the apparatus as an image-formed material (a print or copy).

The surface of the electrophotographic

photosensitive member 1 from which images have been transferred is brought to removal of the toner remaining after the transfer, through a cleaning means 9. Thus, the surface is cleaned. The electrophotographic

5 photosensitive member is further subjected to charge elimination by pre-exposure light 10 emitted from a pre-exposure means (not shown), and then repeatedly used for the formation of images. Incidentally, where the primary charging means 3 is a contact charging means
10 making use of a charging roller or the like as shown in Fig. 4, the pre-exposure is not necessarily required.

In the present invention, the apparatus may be constituted of a combination of plural components integrally combined as a process cartridge from among the
15 constituents such as the above electrophotographic photosensitive member 1, primary charging means 3, developing means 5 and cleaning means 9 so that the process cartridge is detachably mountable on the main body of an electrophotographic apparatus such as a
20 copying machine or a laser beam printer. For example, at least one of the primary charging means 3, the developing means 5 and the cleaning means 9 may integrally be supported in a cartridge together with the electrophotographic photosensitive member 1 to form a
25 process cartridge 21 that is detachably mountable on the main body of the apparatus through a guide means 22 such as rails provided in the main body of the apparatus.

In the case when the electrophotographic apparatus is a copying machine or a printer, the exposure light 4 is light reflected from, or transmitted through, an original, or light irradiated by the scanning of a laser beam, the driving of an LED array or the driving of a liquid-crystal shutter array according to signals obtained by reading an original through a sensor and converting the information into signals.

The electrophotographic photosensitive member of the present invention may be not only applied to electrophotographic copying machines, but also widely applicable in the fields where electrophotography is applied, e.g., laser beam printers, CRT printers, LED printers, liquid-crystal printers, and laser plate making.

The present invention is described below in greater detail by giving Examples. Note, however, that the present invention is by no means limited to these Examples. In the following Examples and Comparative Examples, "part(s)" refers to "part(s) by weight".

Example 1

An aluminum cylinder of 30 mm in diameter and 357 mm in length was coated thereon by dip coating with a coating dispersion made up of the following materials, followed by heat curing at 140°C for 30 minutes to form a conductive layer with a layer thickness of 15 μ m.

Conductive pigment: SnO₂-coated barium sulfate

10 parts

Resistance-adjusting pigment: Titanium oxide 2 parts
Binder resin: Phenol resin 6 parts
Leveling agent: Silicone oil 0.001 part
Solvent: Methanol/methoxypropanol = 2/8 20 parts

5 Next, on the conductive layer, a solution prepared
by dissolving 3 parts of N-methoxymethylated nylon and 3
parts of copolymer nylon in a mixed solvent of 65 parts
of methanol and 30 parts of n-butanol was applied by dip
coating, followed by drying to form an intermediate layer
10 with a layer thickness of 0.7 μm .

Next, 4 parts of hydroxygallium phthalocyanine
having strong peaks at Bragg's angles ($2\theta \pm 0.2^\circ$) of 7.4°
and 28.2° in the $\text{CuK}\alpha$ characteristic X-ray diffraction, 2
parts of polyvinyl butyral resin (trade name: S-LEC BX-1;
15 available from Sekisui Chemical Co., Ltd.) and 60 parts
of cyclohexanone were subjected to dispersion for 4 hours
by means of a sand mill making use of glass beads of 1 mm
in diameter, followed by addition of 100 parts of ethyl
acetate to prepare a charge generation layer coating
20 dispersion. This was coated on the intermediate layer by
dip coating, followed by drying to form a charge
generation layer with a layer thickness of 0.25 μm .

Next, 4 parts of the charge-transporting material
shown as the exemplary compound (CT-1) and 10 parts of
25 polycarbonate resin (repeating structural unit example
(PC-5); trade name: IUPILON Z-400; available from
Mitsubishi Engineering Plastics Co., Ltd.) were dissolved

in a mixed solvent of 80 parts of monochlorobenzene and 20 parts of dimethoxymethane to prepare a charge transport layer coating solution. This was applied on the charge generation layer by dip coating, followed by
5 drying at 120°C for 1 hour to form a charge transport layer with a layer thickness of 25 μm . Thus, an electrophotographic photosensitive member used in Example 1 was produced.

Next, the electrophotographic photosensitive member
10 produced was evaluated as described below.

An evaluation apparatus used was a remodeled machine of a laser beam printer LBP-950, manufactured by CANON INC., (process speed 144.5 mm/sec; contact charging system). The printer was so remodeled that the control of
15 primary charging is changed from constant-current control to constant-voltage control (potential at non-image areas on the photosensitive member: always -680 V). Evaluation was made in a normal temperature and high humidity (23°C, 90%RH) environment.

20 The light-area potential (V_l) of the electrophotographic photosensitive member produced and variations in light-area potential (ΔV_l) were evaluated, setting the light source of the apparatus in such a way that the amount of imagewise exposure light came to be
25 0.5 $\mu\text{J}/\text{cm}^2$ as the amount of light on the electrophotographic photosensitive member surface. To evaluate variations in potential characteristics as a

result of repeated use of the electrophotographic
photosensitive member, images were continuously
reproduced on 10,000 sheets of A4-size plain paper, and
the surface potential before and after measurement. To
5 measure the surface potential of the electrophotographic
photosensitive member, the developing assembly was
changed for a jig which was so fastened that a probe for
measuring the potential was positioned at a position of
180 mm from the upper end of the electrophotographic
10 photosensitive member. The surface potential was measured
at the position of the developing assembly.

Images were also reproduced on 40,000 sheets of
A4-size plain paper in an intermittent mode in which
printing was stopped once for each image reproduction on
15 one sheet. Images were evaluated every 1,000 sheets, and
a point in time at which image quality was seen to
deteriorate (smeared images or scratch images) was
regarded as the limit of running (extensive operation).

After the image reproduction on 40,000 sheets,
20 evaluation was further made on any scratches on the
electrophotographic photosensitive member. To make
evaluation on the scratches, evaluation according to the
ten-point average roughness (Rzjis) evaluation prescribed
in JIS B-0601:2001 was made (evaluation length: 8 mm)
25 using a surface roughness measuring instrument
(SURFCOADER SE-3400, manufactured by Kosaka Laboratory
Ltd.).

The results of evaluation are shown in Table 1.

Examples 2 to 18

Electrophotographic photosensitive members were produced in the same manner as in Example 1 except that
5 charge-transporting materials shown in Table 1 were used as the charge-transporting material. Evaluation was made in the same way. The results are shown in Table 1.

Examples 19 to 25

Electrophotographic photosensitive members were
10 produced in the same manner as in Example 1 except that charge-transporting materials shown in Table 1 were used as the charge-transporting material and a resin shown as the repeating structural unit example (PA-2)
(weight-average molecular weight Mw: 120,000; weight
15 ratio of terephthalic acid to isophthalic acid in resin: terephthalic acid/isophthalic acid = 50/50) was used as the binder resin. Evaluation was made in the same way. The results are shown in Table 1.

Examples 26 to 30

20 Electrophotographic photosensitive members were produced in the same manner as in Example 1 except that charge-transporting materials shown in Table 1 were used as the charge-transporting material and a resin shown as the repeating structural unit example (PA-10)
25 (weight-average molecular weight Mw: 130,000; weight ratio of terephthalic acid to isophthalic acid in resin: terephthalic acid/isophthalic acid = 70/30) was

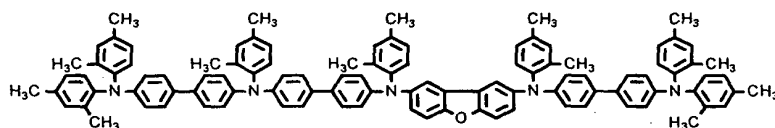
used as the binder resin. Evaluation was made in the same way. The results are shown in Table 1.

Examples 31 to 37

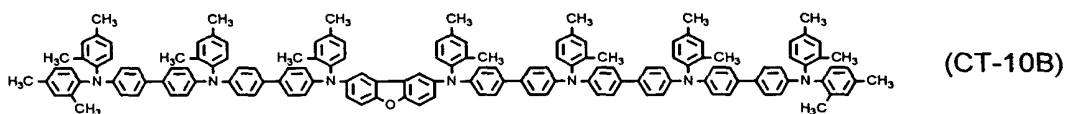
Electrophotographic photosensitive members were produced in the same manner as in Example 1 except that charge-transporting materials shown in Table 1 were used as the charge-transporting material and a copolymer resin having the structure represented by the repeating structural unit examples (PA-2) and (PA-9) (weight-average molecular weight Mw: 125,000; weight ratio of terephthalic acid to isophthalic acid in resin: terephthalic acid/isophthalic acid = 50/50) was used as the binder resin. Evaluation was made in the same way. The results are shown in Table 1.

Example 38

An electrophotographic photosensitive member was produced in the same manner as in Example 1 except that 3.6 parts of the compound represented by the above Formula (CT-10), 0.2 part of a compound represented by the following Formula (CT-10A) and 0.2 part of a compound represented by the following Formula (CT-10B) were used as the charge-transporting material. Evaluation was made in the same way. The results are shown in Table 1.



(CT-10A)



Example 39

An electrophotographic photosensitive member was produced in the same manner as in Example 1 except that

5 3.8 parts of the compound represented by the above Formula (CT-10), 0.1 part of the compound represented by the above Formula (CT-10A) and 0.1 part of the compound represented by the above Formula (CT-10B) were used as the charge-transporting material. Evaluation was made in

10 the same way. The results are shown in Table 1.

Example 40

An electrophotographic photosensitive member was produced in the same manner as in Example 1 except that

3.9 parts of the compound represented by the above

15 Formula (CT-10), 0.05 part of the compound represented by the above Formula (CT-10A) and 0.05 part of the compound represented by the above Formula (CT-10B) were used as the charge-transporting material. Evaluation was made in the same way. The results are shown in Table 1.

20 Example 41

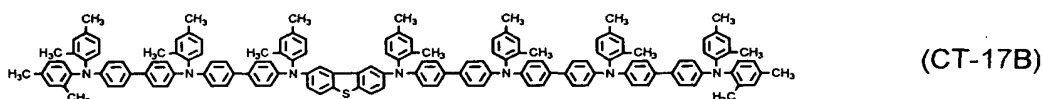
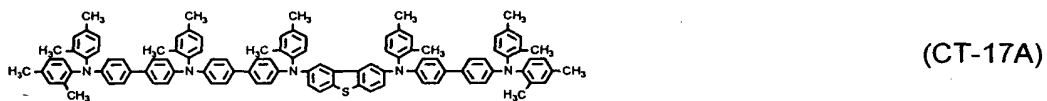
An electrophotographic photosensitive member was produced in the same manner as in Example 1 except that

3.6 parts of the compound represented by the above

Formula (CT-17), 0.2 part of a compound represented by

25 the following Formula (CT-17A) and 0.2 part of a compound represented by the following Formula (CT-17B) were used

as the charge-transporting material. Evaluation was made in the same way. The results are shown in Table 1.



5 Example 42

An electrophotographic photosensitive member was produced in the same manner as in Example 1 except that 3.8 parts of the compound represented by the above Formula (CT-17), 0.1 part of the compound represented by the above Formula (CT-17A) and 0.1 part of the compound represented by the above Formula (CT-17B) were used as the charge-transporting material. Evaluation was made in the same way. The results are shown in Table 1.

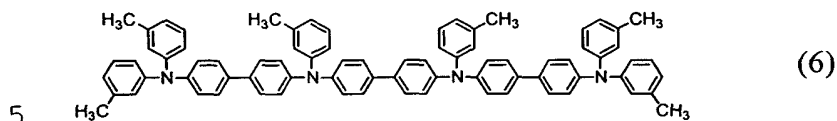
Example 43

15 An electrophotographic photosensitive member was produced in the same manner as in Example 1 except that 3.9 parts of the compound represented by the above Formula (CT-17), 0.05 part of the compound represented by the above Formula (CT-17A) and 0.05 part of the compound represented by the above Formula (CT-17B) were used as the charge-transporting material. Evaluation was made in the same way. The results are shown in Table 1.

Comparative Example 1

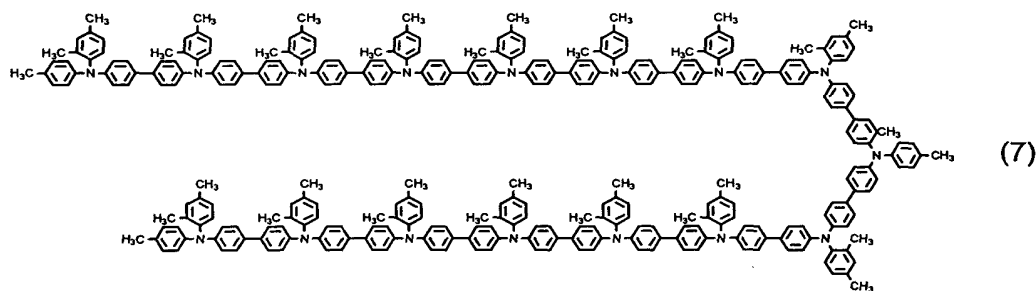
25 An electrophotographic photosensitive member was produced in the same manner as in Example 1 except that a

compound represented by the following Formula (6) was used as the charge-transporting material. Evaluation was made in the same way. The results are shown in Table 1.



Comparative Example 2

An electrophotographic photosensitive member was produced in the same manner as in Example 1 except that a compound represented by the following Formula (7) was used as the charge-transporting material. Evaluation was made in the same way. The results are shown in Table 1.

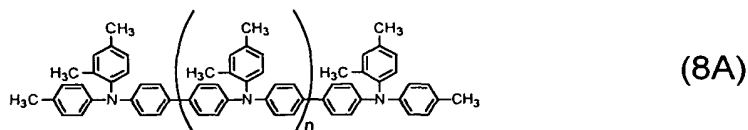


15 Comparative Example 3

A compound represented by the following Formula (8A) was produced by polymerization reaction disclosed in International Publication No. WO00/078843. The compound represented by the following Formula (8A) was a charge-transporting material composed of a mixture of compounds having repeating structural units $n=4:n=5:n=6:n=7:n=8$ in a proportion of 7:23:50:17:3, as

20

found by the peak area ratio in measurement by usual GPC (gel permeation chromatography; column used: KF-802, available from Showa Denko K.K.; development solvent: methanol/tetrahydrofuran = 7/3; detector: IR detector; 5 molecular weight of sample: determined in terms of polystyrene). An electrophotographic photosensitive member was produced in the same manner as in Example 1 except that this compound represented by the following Formula (8A) was used as the charge-transporting material. 10 Evaluation was made in the same way. The results are shown in Table 1.

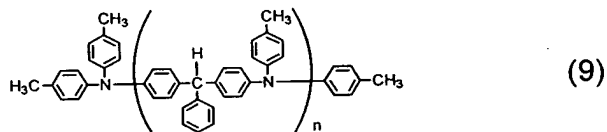


Comparative Example 4

The compound used in Comparative Example 3, 15 represented by Formula (8A), was purified by preparative gel permeation chromatography mainly for the compound the number n of repeating structural unit of which was 6, to obtain a compound (8B) composed of a mixture of compounds having repeating structural units n=5:n=6:n=7 in a 20 proportion of 16:74:10, as found by the peak area ratio in the above measurement by GPC. An electrophotographic photosensitive member was produced in the same manner as in Example 1 except that this compound (8B) was used as the charge-transporting material. Evaluation was made in 25 the same way. The results are shown in Table 1.

Comparative Example 5

A compound represented by the following Formula (9) was produced by the process disclosed in Japanese Patent Application Laid-open No. S61-151545. The compound represented by the following Formula (9) was a charge-transporting material composed of a mixture of compounds having repeating structural units $n=5:n=6:n=7$ in a proportion of 13:68:19, as found by the peak area ratio in the above measurement by GPC. An electrophotographic photosensitive member was produced in the same manner as in Example 1 except that this compound represented by the following Formula (9) was used as the charge-transporting material. Evaluation was made in the same way. The results are shown in Table 1.



Comparative Examples 6 to 10

Electrophotographic photosensitive members were produced in the same manner as in Example 1 except that the charge-transporting materials used in Comparative Examples 1 to 5 were used as the charge-transporting material and the resin used in Example 26 was used as the binder resin. Evaluation was made in the same way. The results are shown in Table 1.

According to the present invention, the use of the electrophotographic photosensitive member described above

makes it possible to provide an electrophotographic
photosensitive member which can form images with good
quality even in repeated use over a long period of time,
as having high mechanical strength, high wear resistance
5 and good electrophotographic performance.

Table 1

Charge-transporting material		* based on total weight of charge-transporting material				
		Proportion of compound(s) * (wt.%)	VL (-V)	ΔV_L (V)	Image evaluation	Rzjis (μm)
Example:						
1	(CT-1)	100	200	-12	No problem after 40,000. (sheets)	1.8
2	(CT-2)	100	198	-12	No problem after 40,000.	1.8
3	(CT-3)	100	198	-13	No problem after 40,000.	1.8
4	(CT-9)	100	200	-8	No problem after 40,000.	1.7
5	(CT-10)	100	198	-8	No problem after 40,000.	1.7
6	(CT-13)	100	203	-12	No problem after 40,000.	1.6
7	(CT-15)	100	203	-12	No problem after 40,000.	1.6
8	(CT-16)	100	198	-8	No problem after 40,000.	1.7
9	(CT-17)	100	198	-8	No problem after 40,000.	1.6
10	(CT-20)	100	200	-10	No problem after 40,000.	1.6
11	(CT-34)	100	203	-12	No problem after 40,000.	1.8
12	(CT-39)	100	198	-15	Smeared i. after 34,000.	1.9
13	(CT-45)	100	205	-13	Smeared i. after 36,000.	1.8
14	(CT-48)	100	200	-10	Smeared i. after 34,000.	1.8
15	(CT-54)	100	200	-15	Smeared i. after 32,000.	1.8
16	(CT-58)	100	215	-15	Smeared i. after 34,000.	1.8
17	(CT-59)	100	210	-14	Smeared i. after 38,000.	1.9
18	(CT-60)	100	212	-15	Smeared i. after 34,000.	1.8
19	(CT-2)	100	203	-7	No problem after 40,000.	1.2
20	(CT-3)	100	203	-8	No problem after 40,000.	1.2

Smeared i.: Smeared images appeared

Table 1 (cont'd)

Charge-transporting material		* based on total weight of charge-transporting material				Rzjis (μm)
Proportion of compound(s) * (wt.%)		V1 (-V)	$\Delta V1$ (V)	Image evaluation		
Example:						
41	(CT-17) : (CT-17A) : (CT-17B)	90:5:5	205	-16	No problem after 40,000.	1.7
42	(CT-17) : (CT-17A) : (CT-17B)	95:2.5:2.5	200	-10	No problem after 40,000.	1.7
43	(CT-17) : (CT-17A) : (CT-17B)	99:0.5:0.5	198	-8	No problem after 40,000.	1.7
Comparative Example:						
1	(6)	100	270	40	Scratch i. after 18,000.	3.8
2	(7)	100	195	-103	Smeared i. after 4,000.	2.0
3	(8A)	7:23:50:17:3	225	-40	Smeared i. after 9,000.	2.3
4	(8B)	16:74:10	223	-25	Smeared i. after 17,000.	2.1
5	(9)	13:68:19	240	-30	Smeared i. after 24,000.	2.3
6	(6)	100	293	27	Scratch i. after 23,000.	3.2
7	(7)	100	203	-140	Smeared i. after 4,000.	2.3
8	(8A)	7:23:50:17:3	245	-70	Smeared i. after 9,000.	2.5
9	(8B)	16:74:10	240	-43	Smeared i. after 14,000.	2.4
10	(9)	13:68:19	250	-50	Smeared i. after 20,000.	2.1

Scratch i.: Scratch images appeared; Smeared i.: Smeared images appeared